

# Chapter 2: Acids, Bases and Salts

Science Worksheet

	Standard: 10th	Subject: Science	
Student Name:			

Q1. Match the chemical substances given in Column (A) with their appropriate application given in Column (B):

	Column (A)		Column (8)	
Α.	Bleaching powder	1.	Preparation of glass	
В.	Baking soda	2.	Production of H <sub>2</sub> and Cl <sub>2</sub>	
C.	Washing soda	3.	Decolourisation	
D.	Sodium chloride	4.	Antacid	

- 1. A-2, B-1, C-4, D-3
- 2. A-3, B-2, C-4, D-1
- 3. A-3, B-4, C-1, D-2
- 4. A-2, B-4, C-1, D-3
- Q2. 10ml, of a solution of NaOH is found to be completely neutralized by 8ml, of a given solution of HCI. If we take 20ml, of same solution of NaOH, the amount of HCI solution required to neutralize it will be:
  - 1. 4mL
  - 2. 8mL
  - 3.12mL
  - 4. 16mt.
- Q3. Calcium phosphate is present in tooth enamel. Its nature is:
  - 1. Basic.
  - 2. Acidic.
  - 3. Neutral.
  - 4. Amphoteric.
- Q4. Which of the following salts will give an aqueous solution having pH of almost 7?
  - 1. NH<sub>a</sub>NO<sub>3</sub>
  - 2. NH<sub>2</sub>CI
  - 3. CaCl<sub>2</sub>.



# **Chapter 1acids Bases Answers**

Stanley E. Manahan

#### **Chapter 1acids Bases Answers:**

**Science For Tenth Class Part 1 Physics** Lakhmir Singh & Manjit Kaur, A series of six books for Classes IX and X according to the CBSE syllabus Each class divided into 3 parts Part 1 Physics Part 2 Chemistry Part 3 Biology

Fundamentals of Sustainable Chemical Science Stanley E. Manahan, 2009-03-10 Written by Stanley Manahan Fundamentals of Sustainable Chemical Science has been carefully designed to provide a basic introduction to chemistry including organic chemistry and biochemistry for readers with little or no prior background in the subject Manahan bestselling author of many environmental texts presents the material in a practical Fundamentals of Environmental Chemistry, Third Edition Stanley E. Manahan, 2011-03-05 Written by an expert using the same approach that made the previous two editions so successful Fundamentals of Environmental Chemistry Third Edition expands the scope of book to include the strongly emerging areas broadly described as sustainability science and technology including green chemistry and industrial ecology The new edition includes Increased emphasis on the applied aspects of environmental chemistry Hot topics such as global warming and biomass energy Integration of green chemistry and sustainability concepts throughout the text More and updated questions and answers including some that require Internet research Lecturers Pack on CD ROM with solutions manual PowerPoint presentations and chapter figures available upon qualifying course adoptions The book provides a basic course in chemical science including the fundamentals of organic chemistry and biochemistry. The author uses real life examples from environmetral chemistry green chemistry and related areas while maintaining brevity and simplicity in his explanation of concepts Building on this foundation the book covers environmental chemistry broadly defined to include sustainability aspects green chemistry industrial ecology and related areas These chapters are organized around the five environmental spheres the hydrosphere atmosphere geosphere biosphere and the anthrosphere The last two chapters discuss analytical chemistry and its relevance to environmental chemistry Manahan's clear concise and readable style makes the information accessible regardless of the readers level of chemistry knowledge He demystifies the material for those who need the basics of chemical science for their trade profession or study curriculum as well as for readers who want to have an understanding of the fundamentals of sustainable chemistry in its crucial role in maintaining a livable planet Saraswati **Chemistry Class 10** RP Manchanda, A text book on Chemistry Quick revision to Fundaments of chemistry 10 Amar Bhutani, Quick Revision to Fundamentals of Chemistry Class 10 is a fully exam focused guide crafted to make ICSE Chemistry simple scoring and revision friendly Carefully mapped to the latest syllabus this book provides concept summaries definitions solved numericals balanced equations diagrams and past exam style questions helping students revise quickly and perform with confidence All important chapters Periodic Table Chemical Bonding Acids Bases Salts Metallurgy Organic Chemistry Mole Concept Electrolysis are presented through point wise notes and easy explanations that save time and strengthen retention Features include All textbook questions fully solved Additional ICSE style practice questions Important

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