

- Specific Nature** - Urinase catalyses the hydrolysis of urea and not methyl urea, so these are specific in nature.
- Optimum Temperature** - It is active at 20-30°C
- pH of medium** - It is about 7, for pepsin 1.5-2.5, for trypsin 7.5-8.5
- Concentration** - Dilute solutions are more effective
- Amount of enzyme** - Very small amount can accelerate the reaction
- Enzyme inhibitors** - These compound inhibit the enzyme action, with the help of such compounds, the reaction can be controlled.

HORMONES These are the chemical substances which are produced by endocrine (ductless) glands in the body. Hormones act as chemical messengers.

Some examples of ductless (endocrine) glands are thyroid, pituitary, adrenal, pancreas, testes and ovaries.

Hormones are divided into three types:

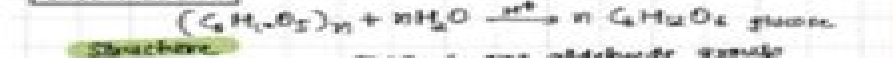
- Steroids
- Peptides
- Amines

Name of endocrine gland	Hormones secreted
Testis	Testosterone
Ovary	Estrogen
Adrenal	Adrenaline
Thyroid glands	Thyroxine
Pituitary gland	FSH
Pancreas	Insulin

Carbohydrates These are optically active polyhydroxy aldehydes / ketones or the substance which give these on hydrolysis are called carbohydrates.

- Classification on the basis of Hydrolysis**
- Mono-saccharides** - can not be hydrolyzed further e.g. glucose, fructose.
 - Oligo-saccharides** - give 2-10 molecules of mono-saccharides e.g. sucrose, maltose, raffinose.
 - Polysaccharides** - give large no. of mono-saccharides e.g. starch, cellulose.

Preparation of Glucose:



Structure

CHO ← one aldehyde group

(CHOH)₄ ← four 2° alcohol

CH₂OH ← one 1° alcohol

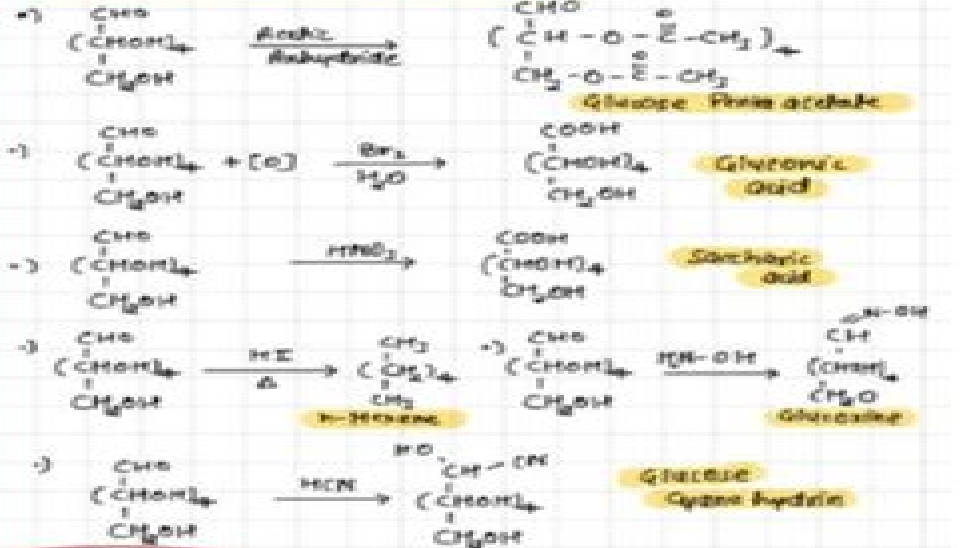


Str. of fructose: { The C, 6 known as anomeric carbon and the other carbons are called anomers }



- Reducing Sugar**
- Free Aldehydic or Ketonic group
 - Reduce Fehling Solⁿ or Tollen's Reagent
 - e.g. Maltose & Fructose
- Non Reducing Sugar**
- Do not have any free aldehydic or ketonic group.
 - Do not reduce Fehling's Reagent and Fehling Solⁿ e.g. Sucrose

Chemical Properties of Glucose



Mutarotation when either of the two forms of glucose is dissolved in water, there is a spontaneous change in specific rotation till the equilibrium value of +52.5°. This is known as mutarotation.

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