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**Edexcel GCE
Core Mathematics C2
Advanced Subsidiary**

Friday 24 May 2013 – Morning

Time: 1 hour 30 minutes

DISCUSSION—A recent study by the American Cancer Society¹ estimated that 1.2 million people in the United States have been exposed to asbestos in the workplace. The authors of this study estimated that 1.2 million people in the United States have been exposed to asbestos in the workplace. The authors of this study estimated that 1.2 million people in the United States have been exposed to asbestos in the workplace.

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¹ *Journal of Leadership Studies*, 1996, 9, 1.

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Materials required for examination

Mathematical Formulas (Click)

Items included with question papers

Candidates may use any calculator allowed by the regulations of the Joint Council for Qualifications. Calculators must not have the facility for symbolic algebra manipulation or symbolic differentiation/integration, or have retrievable mathematical formulae stored in them.

Instructions to Candidates

In the boxes above, write your centre number, candidate number, your surname, initials and signature. Check that you have the correct question paper.

Answer ALL the questions.

You must write your answer for each question in the space following the question.

When a calculator is used, the answer should be given to an appropriate degree of accuracy.

Information for Candidates

A booklet "Mathematical Formulae and Statistical Tables" is provided.

Full marks may be obtained for answers to ALL questions

The marks for individual questions and the parts of questions are shown in round brackets: e.g. (2).

There are 10 questions in this question paper. The total mark for this paper is 75.

There are 32 pages in this question paper. Any blank pages are indicated.

Advice to Candidates

You must ensure that your answers to parts of questions are clearly labelled

You should show sufficient working to make your methods clear to the Examiner.

Answers without working may not gain full credit

[illegible]**TABLE 1**

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web the volume of 1 00 mol 1 00 mol of any gas at stp standard temperature 273 15 k and pressure 1 atm is measured to be 22 414 l 22 414 l we can substitute 101 325 kpa 101 325 kpa for pressure 22 414 l 22 414 l for volume and 273 15 k 273 15 k for temperature into the ideal gas equation and solve for r

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web may 20 2018 french physicist jacques charles 1746 1823 studied the effect of temperature on the volume of a gas at constant pressure charles s law states that the volume of a given mass of gas varies directly with the absolute temperature of the gas when pressure is kept constant the absolute temperature is temperature measured

9 2 relating pressure volume amount and temperature the - Mar 03 2023

web chemists sometimes make comparisons against a standard temperature and pressure stp for reporting properties of gases 273 15 k and 1 atm 101 325 kpa 1 at stp one mole of an ideal gas has a volume of about 22 4 l this is referred to as the standard molar volume figure 9 18

lab pressure lab reports lab pressure volume and temperature - Oct 10 2023

web this lab explores the relationships between pressure volume and temperature in an ideal gas the ideal gas law $pV = nRT$ is used and with it we can explore it by seeing how the pressure and volumes changed in the tubes each time that we

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web figure page index 10 since the number of moles in a given volume of gas varies with pressure and temperature changes

chemists use standard temperature and pressure 273 15 k and 1 atm or 101 325 kpa to report properties of gases

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web pressure volume and temperature discovery of gas laws experiment 5 10 3 19 jared caviglia section 582 ian nicholson

introduction the ideal gas law is a commonly used formula in chemistry it relates pressure temperature and volume this

experiment seeks to prove those relationships by measuring the pressure of gas at various temperatures

pressure volume temperature lab ku school of engineering - Jun 25 2022

web the pressure volume temperature pvt fluids lab in ku s chemical petroleum engineering department is used to study the properties of fluids under a wide range of pressure and temperature

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web volume pressure kpa linear pressure kpa volume ml pressure kpa in part b we observed the relationship between

temperature and pressure the pressure of the 125 ml flask was at 101 36 kpa when the temperature was at 23 2 c

pressure temperature relationship in gases lab studocu - Nov 30 2022

web pressure increases even though there were errors in the experiment the goal of the experiment was met the goal of

declaring what type of relationship it is was clear using the data pressure kpa temperature c temperature k constant k p t 95

82 kpa 16 0 c 289 k 0 33 97 90 kpa 19 0 c 292 k 0 34 98 97 kpa 99 22 kpa 101 4 kpa 19 1 c

pressure volume temperature pvt laboratory department - Sep 09 2023

web aug 13 2021 pressure volume temperature pvt laboratory analysis of fluid properties such as api gravity viscosity

surface and interfacial tension ph refractive index and vapor pressure is performed in the pvt laboratory pressure volume and

temperature relationships of hydrocarbons are studied at the graduate level

che144 thermodynamics engineering lab report saturation vapor pressure - Jul 27 2022

web the properties of water at constant volume can be represented as a function of pressure and temperature as shown in

the diagram below figure 1 the pressure temperature relationship at constant volume the saturation point of water is the

condition at which a phase change occurs from liquid to vapour or vapour to liquid

6 3 relationships among pressure temperature volume and - Jul 07 2023

web figure pageindex 5 the empirically determined relationships among pressure volume temperature and amount of a gas

the thermometer and pressure gauge indicate the temperature and the pressure qualitatively the level in the flask indicates

the volume and the number of particles in each flask indicates relative amounts