

SOLUTIONS AND SOLUBILITY

IGCSE EDEXCEL CHEMISTRY

Lecture & Worksheet

SOLUTIONS AND SOLUBILITY

IGCSE CHEMISTRY

PART 1: PRINCIPLES OF CHEMISTRY

1. Understand how the results of experiments involving the dilution of coloured solutions and diffusion of gases can be explained.

2. Understand the meanings of the terms: solvent, solute, solution, and saturated solution.

3. Match the words to their definitions. (20)

Word	Definition
Solute	A substance dissolved.
Solvent	A liquid which dissolves a solute and forms a liquid.
Solution	A mixture of a solute and a solvent.
Saturated	A solution in which no more solute can be dissolved at a given temperature.
Dilute	A solution containing a small amount of solute.
Concentrated	A solution containing a large amount of solute.
Diffusion	The movement of particles from an area of high concentration to an area of low concentration.
Evaporation	The process by which a liquid changes into a gas.
Crystallisation	The process by which a solid forms from a solution.

4. Write down the names of:

- 1. A solid that dissolves easily. (20)
- 2. A solid that does not dissolve in water. (20)

5. Name of gases which:

- 1. Dissolve easily in water. (20)
- 2. Do not dissolve in water. (20)

6. Name of the solvent in:

- 1. A solution of sugar in water. (20)
- 2. A solution of salt in water. (20)

7. Name of the solute in:

- 1. A solution of sugar in water. (20)
- 2. A solution of salt in water. (20)

8. Name of the solvent in:

- 1. A solution of sugar in water. (20)
- 2. A solution of salt in water. (20)

9. Name of the solute in:

- 1. A solution of sugar in water. (20)
- 2. A solution of salt in water. (20)

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Specification Points

1.3 Understand how the results of experiments involving the dilution of coloured solutions and diffusion of gases can be explained.

1.4 Understand the meanings of the terms: solvent, solute, solution, and saturated solution.

FREQUENCY IN ASSESSMENT

LOW MEDIUM HIGH

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R Barnett



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