

# Introduction to Chemistry

## Section 1.1 The Stories of Two Chemicals

In your textbook, read about the ozone layer.

Use each of the boxes below just once to complete the passage.

|              |             |                       |            |
|--------------|-------------|-----------------------|------------|
| atmosphere   | integrated  | ozone                 | ozone hole |
| stratosphere | troposphere | ultraviolet radiation |            |

Earth's (90) \_\_\_\_\_ is made up of several layers. The air we breathe makes up the lowest layer. This layer is called the (80) \_\_\_\_\_. The next layer up is called the (82) \_\_\_\_\_. This layer contains a protective (81) \_\_\_\_\_ layer.

Closest to us is the (83) \_\_\_\_\_, made by ultraviolet radiation in the upper part of the stratosphere. The ozone hole is a layer around Earth, with a (84) \_\_\_\_\_ without ozone, you are more likely to get a sunburn or possibly skin cancer. The covering of the ozone layer is called the (85) \_\_\_\_\_. In developed nations, without ozone will increase on Earth, air called to have three or four (86) \_\_\_\_\_.

In your textbook, read about chlorofluorocarbons.

Use each statement below, writing in or below.

- \_\_\_\_\_ 8. CFCs is another name for a chlorofluorocarbon.
- \_\_\_\_\_ 9. CFCs are made up of carbon, fluorine, and chlorine.
- \_\_\_\_\_ 10. All CFCs are synthetic chemicals.
- \_\_\_\_\_ 11. CFCs usually react easily with other chemicals.
- \_\_\_\_\_ 12. CFCs were designed as replacements for toxic refrigerants.

# Chemistry Study Guide Content Mastery Teacher

**Charles Riborg Mann**



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