



Chemical Reactions

Name _____

Date _____ Pd _____

Score _____ / 50 points

Instructions: Select the answer that most appropriately answers the question.

1. Which of the following is NOT a sign of a chemical reaction?
 - a. Gas Formation
 - b. Shape Change
 - c. Production of Light
 - d. Change in Smell or Taste
 - e. Endothermic Reaction
2. Which of the following is a chemical reaction?
 - a. Shredding a piece of paper
 - b. Ice melting on a hot day
 - c. Hair bleached white
 - d. Sugar dissolved in water
3. You find a half-eaten banana in your brother's lunch box; it has turned brown and smells funny. What sign of a chemical reaction has taken place?
 - a. Gas Formation
 - b. Production of Light
 - c. Change in Smell or Taste
 - d. Endothermic Reaction
4. In the reaction $\text{CaCO}_3 \rightarrow \text{CaO} + \text{CO}_2$, what are the reactants?
 - a. CaCO_3 and CaO
 - b. CaCO_3
 - c. CaO and CO_2
 - d. CO_2
5. **True or False:** There will never be more than one sign of a physical change that indicates a chemical reaction has taken place for any given chemical reaction.
 - a. True
 - b. False
6. What is used to represent the elements and compounds in a chemical reaction?
 - a. Element symbols from the periodic table
 - b. Full name of the element or compound
 - c. Three letter abbreviations of the element or compound

Chemistry Unit 6 Test Chemical Reactions

Henry George Denham



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