

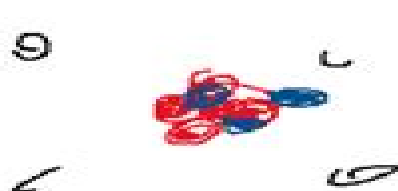
Chemthink guided notes – Ions

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At certain times atoms can gain or lose one or more ____electrons_____. When this happens we say that the atom has formed an _ion_____. Ions are just individual atoms or groups of atoms that has a charge. This charge comes from an imbalance of in the number of ____positive_____ protons and ____negative_____ electrons.

Since the electrons don't contribute to the ____mass_____ of the atom, gaining or losing electrons does not change the overall mass of the atom. It does affect it in other ways. Ions can behave very differently than the atoms from which they were formed. As we will see in later units, forming ions is one way that atoms bond together to make new substances.

Draw a picture of the lithium atom and write the particles in the particle inventory



Protons 3

Electrons 3

Neutrons 4 _____

total charge = 0

symbol: Li

What happens when an electron is lost?

symbol?

Draw a picture of the oxygen atom and write the particles in the particle inventory



Protons 8

Electrons 8

Neutrons 8 _____

total charge = 0

symbol: O

What happens when two electrons are gained?

An ion of -2 charge forms

symbol? O^{2-}

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VM Jensen

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the VM Jensen text.

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