

Intro. to Chemical Reactions

Name _____

Date _____

Class _____

Formula Review

1. Write the formulas for the following compounds:

- a. aluminum phosphate _____ b. sodium phosphate _____
 c. copper (II) phosphate _____ d. sodium sulfate _____

2. Write the formulas for the following compounds:

- a. CaCl_2 _____ b. FeCl_3 _____
 c. FeCl_2 _____ d. CuCl_2 _____

Chemical Reactions: Virtual Energy Unit 10

Write down observations for the reaction as it occurs:



There is no observable gas or color change when the two solutions are mixed.

Experiment 10 involves the reaction of hydrochloric acid with solid sodium hydroxide.

Write a balanced chemical equation for the reaction between sodium hydroxide and hydrochloric acid. Label the phases for each.

Chemical Reaction Classification

3. Classify the following reaction as one of the types that you have studied:



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James Jeans



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F&S Index International Annual ,1997

An Introduction to the Kinetic Theory of Gases James Jeans,1962

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