

Exercises (100)

Controlling Chemical Reactions

If the statement is true, write "true". If the statement does change the statement word or words to make the statement true.

1. Le Chatelier's Principle summarizes the response of a system to a change in equilibrium.
2. True the concentration of a reactant in a closed system is the equilibrium of the reaction.
3. Le Chatelier's Principle states that the effect of a change in a chemical equilibrium.
4. Le Chatelier's Principle states that a change in equilibrium will shift the reaction to the right or left.
5. Le Chatelier's Principle states that a change in equilibrium will shift the reaction to the right or left.

Put in the blank a complete word or statement.

1. The reaction of a chemical system at equilibrium will shift to the right if the concentration of the reactants is increased.
2. In an equilibrium system, the concentration of the reactants is increased if the concentration of the products is decreased.
3. Increasing the concentration of the reactants in an equilibrium system will shift the reaction to the right.
4. The reaction of a chemical system at equilibrium will shift to the right if the concentration of the reactants is increased.
5. The reaction of a chemical system at equilibrium will shift to the right if the concentration of the reactants is increased.

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