

20.1**DESCRIBING ACIDS AND BASES****SECTION REVIEW****Objectives**

- List the properties of acids and bases.
- Name an acid or base when given the formula.

Key Terms

- acid
- base

Part A Completion

After this completion exercise is done, your understanding of the concepts and terms is clear and reinforced in this section. Each blank has the compound with a name, about phrase, or a symbol.

Aqueous solutions of _____ 1 _____ are electrolytes. Acids react with many _____ 2 _____ to produce _____ 3 _____ gas. _____ 4 _____ are compounds that react with acids to form _____ 5 _____ and a salt.

Acids produce _____ 6 _____ ions when dissolved in water. There is a standard system for naming acids. For example, HBr is called _____ 7 _____ acid, while HNO_3 is called _____ 8 _____ acid.

Bases produce _____ 9 _____ ions when dissolved in water. Some compounds that are _____ 10 _____ are named in the same way as other ionic compounds.

1. _____

2. _____

3. _____

4. _____

5. _____

6. _____

7. _____

8. _____

9. _____

10. _____

Part B True-False

Circlely write *if true* or *if false* in the blank after the *if*, *AC* for *always true*, *SC* for *sometimes true*, *FC* for *false*.

- _____ 11. A compound that contains hydrogen atoms will be an acid when dissolved in water.
- _____ 12. An acid contains one or more hydroxide ions.
- _____ 13. Chemists have a special system for naming bases.

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Fluid and Electrolyte Therapy, An Issue of Veterinary Clinics of North America: Food Animal Practice Geof W. Smith,2014-07-28 Fluid and Electrolyte Therapy An Issue of Veterinary Clinics of North America Food Animal Practice

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