

Guided Practice: Balancing Equations & Classifying Chemical Reactions

Evidence of a Chemical Formula

10) List the four clues which indicate a chemical reaction has occurred.

-
-
-
-

20) Define chemical reaction.

30) For each of the following determine whether or not a chemical reaction occurred by stating yes if a chemical reaction occurred or no if a chemical reaction did not occur.

- Roasting a marshmallow
- Rotting of fruit
- Sugar dissolving in water
- Food coloring added to water
- Mixing of two solutions and bubbles are produced

Chemical Equations

40) Compare and contrast chemical reaction and chemical equation.

50) Use the equation below to answer the following questions:



- True or False - Mg and H₂O are the products.
- What does the (l) after H₂O stand for?
- What is the physical state of H₂?
- List one of the reactants.

60) Identify the reactants and products and write the unbalanced equation (including symbols for the physical states) for the following chemical reaction.

Gaseous ammonia reacts with gaseous oxygen to form gaseous nitrogen monoxide and gaseous water.

Balancing Chemical Equations

70) True or False - In a chemical reaction, all atoms present in the reactants must be accounted for in the products.

Describing Chemical Reactions Answer Key Guided Practice

Hussin A.Rothana



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