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Date Submitted: 9/25/06
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Determination of a Chemical Formula

Objective:

To determine the percentage composition and simplest formula for an oxide of copper by its reduction with methane gas (CH_4) at approximately 500°C . This experiment is a study of stoichiometry.

Experimental Data:

Mass of Test Tube	$17.726 \pm .001 \text{ g}$
Mass of Test Tube and Cu_2O	$19.705 \pm .001 \text{ g}$
Mass of Cu_2O	$1.979 \pm .001 \text{ g}$
Mass of Test Tube and Cu	$19.313 \pm .001 \text{ g}$
Mass of Cu	$1.587 \pm .001 \text{ g}$
Mass of O	$0.392 \pm .001 \text{ g}$

Sample Calculation:

Mass percentage calculation:

$$\frac{\text{Mass of Atom}}{\text{Mass of Compound}} \times 100 = \frac{1.587 \pm .001 \text{ g of Cu}}{1.979 \pm .001 \text{ g of Cu}_2\text{O}} \times 100 = 79.73\%$$

Calculation of a Chemical Formula:

Assume a 100 g compound and calculate individual mass using mass percentage.

$$79.73 \text{ g of Cu} \times \frac{1 \text{ mol of Cu}}{63.55 \text{ g per mol}} = 1.255 \text{ mol of Cu}$$

$$20.07 \text{ g of O} \times \frac{1 \text{ mol of O}}{16.00 \text{ g per mol}} = 1.254 \text{ mol of O}$$

Results and Conclusions:

Mass of Test Tube	$17.726 \pm .001 \text{ g}$
Mass of Test Tube and Cu_2O	$19.705 \pm .001 \text{ g}$
Mass of Cu_2O	$1.979 \pm .001 \text{ g}$
Mass of Test Tube and Cu	$19.313 \pm .001 \text{ g}$
Mass of Cu	$1.587 \pm .001 \text{ g}$
Mass of O	$0.392 \pm .001 \text{ g}$
Mass Percentage of Cu	79.73%
Mass Percentage of O	20.07%

Cu and O have a one to one mole ratio; therefore the empirical formula is CuO .

Determination Of A Chemical Formula Lab Report

Chunlong Zhang



Determination Of A Chemical Formula Lab Report:

Fundamentals of Environmental Sampling and Analysis Chunlong Zhang, 2007-03-09 An integrated approach to understanding the principles of sampling chemical analysis and instrumentation This unique reference focuses on the overall framework and why various methodologies are used in environmental sampling and analysis An understanding of the underlying theories and principles empowers environmental professionals to select and adapt the proper sampling and analytical protocols for specific contaminants as well as for specific project applications Covering both field sampling and laboratory analysis Fundamentals of Environmental Sampling and Analysis includes A review of the basic analytical and organic chemistry statistics hydrogeology and environmental regulations relevant to sampling and analysis An overview of the fundamentals of environmental sampling design sampling techniques and quality assurance quality control QA QC essential to acquire quality environmental data A detailed discussion of the theories of absorption spectroscopy for qualitative and quantitative environmental analysis metal analysis using various atomic absorption and emission spectrometric methods and the instrumental principles of common chromatographic and electrochemical methods An introduction to advanced analytical techniques including various hyphenated mass spectrometries and nuclear magnetic resonance spectroscopy With real life case studies that illustrate the principles plus problems and questions at the end of each chapter to solidify understanding this is a practical hands on reference for practitioners and a great textbook for upper level undergraduates and graduate students in environmental science and engineering English in Analytical Chemistry. Communicating about Methods & Techniques. Книга для студента Надежда Зорина, Александра Соболева, 2022-10-19

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Table of Contents Determination Of A Chemical Formula Lab Report

1. Understanding the eBook Determination Of A Chemical Formula Lab Report
 - The Rise of Digital Reading Determination Of A Chemical Formula Lab Report
 - Advantages of eBooks Over Traditional Books
2. Identifying Determination Of A Chemical Formula Lab Report
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Determination Of A Chemical Formula Lab Report
 - User-Friendly Interface
4. Exploring eBook Recommendations from Determination Of A Chemical Formula Lab Report
 - Personalized Recommendations
 - Determination Of A Chemical Formula Lab Report User Reviews and Ratings
 - Determination Of A Chemical Formula Lab Report and Bestseller Lists
5. Accessing Determination Of A Chemical Formula Lab Report Free and Paid eBooks
 - Determination Of A Chemical Formula Lab Report Public Domain eBooks
 - Determination Of A Chemical Formula Lab Report eBook Subscription Services
 - Determination Of A Chemical Formula Lab Report Budget-Friendly Options
6. Navigating Determination Of A Chemical Formula Lab Report eBook Formats
 - ePub, PDF, MOBI, and More
 - Determination Of A Chemical Formula Lab Report Compatibility with Devices
 - Determination Of A Chemical Formula Lab Report Enhanced eBook Features
7. Enhancing Your Reading Experience

- Adjustable Fonts and Text Sizes of Determination Of A Chemical Formula Lab Report
 - Highlighting and Note-Taking Determination Of A Chemical Formula Lab Report
 - Interactive Elements Determination Of A Chemical Formula Lab Report
8. Staying Engaged with Determination Of A Chemical Formula Lab Report
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Determination Of A Chemical Formula Lab Report
 9. Balancing eBooks and Physical Books Determination Of A Chemical Formula Lab Report
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Determination Of A Chemical Formula Lab Report
 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
 11. Cultivating a Reading Routine Determination Of A Chemical Formula Lab Report
 - Setting Reading Goals Determination Of A Chemical Formula Lab Report
 - Carving Out Dedicated Reading Time
 12. Sourcing Reliable Information of Determination Of A Chemical Formula Lab Report
 - Fact-Checking eBook Content of Determination Of A Chemical Formula Lab Report
 - Distinguishing Credible Sources
 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
 14. Embracing eBook Trends
 - Integration of Multimedia Elements
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