

The background of the cover is a photograph of laboratory glassware. In the center, a 250 mL Erlenmeyer flask is filled with a clear liquid. To its left is a round-bottom flask, and to its right is a graduated cylinder. In the foreground, a portion of a periodic table of elements is visible, showing the bottom half of the table. The entire scene is bathed in a blue light, giving it a cool, scientific feel.

Sixth Edition

Descriptive Inorganic Chemistry

**Geoff Rayner-Canham
Tina Overton**

Descriptive Inorganic Chemistry 6th Edition

Nikolay Gerasimchuk



Descriptive Inorganic Chemistry 6th Edition:

Descriptive Inorganic Chemistry James E. House, Kathleen A. House, 2015-09-10 House's *Descriptive Inorganic Chemistry* Third Edition provides thoroughly updated coverage of the synthesis reactions and properties of elements and inorganic compounds. Ideal for the one semester ACS recommended sophomore or junior level course in descriptive inorganic chemistry, this resource offers a readable and engaging survey of the broad spectrum of topics that deal with the preparation, properties, and use of inorganic materials. Using rich graphics to enhance content and maximize learning, the book covers the chemical behavior of the elements, acid-base chemistry, coordination chemistry, organometallic compounds, and numerous other topics to provide a coherent treatment of the field. The book pays special attention to key subjects such as chemical bonding and Buckminsterfullerenes and includes new and expanded coverage of active areas of research such as bioinorganic chemistry, green chemistry, redox chemistry, nanostructures, and more. Highlights the Earth's crust as the source of most inorganic compounds and explains the transformations of those compounds into useful products. Provides a coherent treatment of the field covering the chemical behavior of the elements, acid-base chemistry, coordination chemistry, and organometallic compounds. Connects key topics to real world industrial applications such as in the area of nanostructures. Includes expanded coverage on bioinorganic chemistry, green chemistry, redox chemistry, superacids, catalysis, and other areas of recent development.

Descriptive Inorganic Chemistry Geoff Rayner-Canham, Tina Overton, 2014-09-19 This bestselling text gives students a less rigorous, less mathematical way of learning inorganic chemistry using the periodic table as a context for exploring chemical properties and uncovering relationships between elements in different groups. The authors help students understand the relevance of the subject to their lives by covering both the historical development and fascinating contemporary applications of inorganic chemistry, especially in regard to industrial processes and environmental issues. The new edition offers new study tools, expanded coverage of biological applications, and new help with problem solving.

Arrow Pushing in Inorganic Chemistry Abhik Ghosh, Steffen Berg, 2014-07-25 Involved as it is with 95% of the periodic table, inorganic chemistry is one of the foundational subjects of scientific study. Inorganic catalysts are used in crucial industrial processes, and the field to a significant extent also forms the basis of nanotechnology. Unfortunately, the subject is not a popular one for undergraduates. This book aims to take a step to change this state of affairs by presenting a mechanistic, logical introduction to the subject. Organic teaching places heavy emphasis on reaction mechanisms, arrow pushing, and the authors of this book have found that a mechanistic approach works just as well for elementary inorganic chemistry. As opposed to listening to formal lectures or learning the material by heart by teaching students to recognize common inorganic species as electrophiles and nucleophiles, coupled with organic style arrow pushing, this book serves as a gentle and stimulating introduction to inorganic chemistry, providing students with the knowledge and opportunity to solve inorganic reaction mechanisms. The first book to apply the arrow pushing method to inorganic chemistry teaching. With the

reaction mechanisms approach arrow pushing students will no longer have to rely on memorization as a device for learning this subject but will instead have a logical foundation for this area of study Teaches students to recognize common inorganic species as electrophiles and nucleophiles coupled with organic style arrow pushing Provides a degree of integration with what students learn in organic chemistry facilitating learning of this subject Serves as an invaluable companion to any introductory inorganic chemistry textbook *Descriptive Inorganic Chemistry* J. E. House, Kathleen Ann House, 2001 This title on inorganic chemistry is intended for chemistry biology and earth science students and encompasses theoretical as well as synthetic studies It has relevance for geologists engineers and materials science students

Handbook of Inorganic Compounds Dale L. Perry, 2016-04-19 For anyone that needs property data for compounds CASRN numbers for computer or other searches a consistent tabulation of molecular weights to synthesize inorganic materials on a laboratory scale or information on commercial and other uses for various compounds this volume is the perfect reference This second edition is fully revised and updated New data include optical inorganics radiation detection inorganics thermochromic compounds piezochromic compounds metal ion coordination complexes expanded crystallographic and structural data for inorganics catalysts superconductors and luminescent fluorescent and phosphorescent inorganics

Chemical Literacy and Writing Chemical Reactions Nikolay Gerasimchuk, 2022-09-12 Writing chemical reactions in general and inorganic chemistry is not a trivial task However writing reactions for chemical processes correctly is a clear indicator of proficiency and competence in a subject Unfortunately very few students grasp the concept of the correct writing of chemical reactions quickly and so are unable to move through topics of general analytical and inorganic chemistry freely Because the ability to write and balance different types of chemical reactions is a fundamental issue this becomes a key question of chemical literacy The successful writing of chemical reactions includes two components the prediction of products of these reactions and their possible variations and balancing these reactions providing a material balance between starting compounds and reactions products This book explores that element of the teaching of the fundamentals of chemical literacy writing complete equations of chemical reactions and balancing them It contains 49 figures 22 schemes and 12 tables and 93 problems with answers This book will be very useful for high school students interested in chemical sciences higher education teachers students in colleges and universities majoring in chemistry and biochemistry and chemistry professional working in industry It also contains information about properties of the most common elements and applications of a variety of their chemical compounds

Mendeleev to Oganesson Eric Scerri, Guillermo Restrepo, 2018-02-13 Since 1969 the international chemistry community has only held conferences on the topic of the Periodic Table three times and the 2012 conference in Cusco Peru was the first in almost a decade The conference was highly interdisciplinary featuring papers on geology physics mathematical and theoretical chemistry the history and philosophy of chemistry and chemical education from the most reputable Periodic Table scholars across the world Eric Scerri and Guillermo Restrepo have collected fifteen of the strongest

papers presented at this conference from the most notable Periodic Table scholars The collected volume will contain pieces on chemistry philosophy of science applied mathematics and science education

Periodic Table, The: Past, Present, And Future Geoff Rayner-canham, 2020-08-04 This is an absolutely wonderful book that is full of gems about the elements and the periodic table All in all the book is highly recommended to philosophers of chemistry As philosophers we have a natural tendency to concentrate on generalities and not to get too involved in the specifics and the details Above all else this new book reminds us that such an approach needs to be tempered by a detailed knowledge of the exceptions and features that go against the simplified generalities which we so cherish Read Full Review Eric Scerri Foundations of Chemistry Many questions are dealt with in a clearly written way in this stimulating and innovative book The reader will quickly become interested in the subject and will be taken on tour through this Periodic Table in a very readable way both for students and teachers The number of illustrations is good and clear This book is indeed unique and quite thought provoking This book is highly recommended for students teachers researchers and not only chemists Geologists biochemist and also physicists will find it very interesting to read Read Full Review Chemistry International That fossilized chart on every classroom wall isn't that The Periodic Table Isn't that what Mendel ever devised about a century ago No and No There are many ways of organizing the chemical elements some of which are thought provoking and which reveal philosophical challenges Where does hydrogen belong Can an element occupy more than one location on the chart Which are the Group 3 elements Is aluminum in the wrong place Why is silver like thallium Why is vanadium like molybdenum Why does gold form an auride ion like a halide ion Does an atom know if it is a non metal or metal Which elements are the metalloids Which are the triels So many questions In this stimulating and innovative book the Reader will be taken on a voyage from the past to the present to the future of the Periodic Table This book is unique This book is readable This book is thought provoking It is a multi dimensional examination of patterns and trends among the chemical elements Every reader will discover something about the chemical elements which will provoke thought and a new appreciation as to how the elements relate together

ADVANCED INORGANIC CHEMISTRY, 6TH ED Cotton, 2007-08 Special Features Systematically covers the periodic table and encompasses the chemistry of all chemical elements and their compounds including interpretative discussion in light of the advances in structural chemistry general valence theory and ligand field theory Increases coverage of descriptive chemistry About The Book For more than a quarter century Cotton and Wilkinson's Advanced Inorganic Chemistry has been the source that students and professional chemists have turned to for the background needed to understand current research literature in inorganic chemistry and aspects of organometallic chemistry Like its predecessors this updated Sixth Edition is organized around the periodic table of elements and provides a systematic treatment of the chemistry of all chemical elements and their compounds It incorporates important recent developments with an emphasis on advances in the interpretation of structure bonding and reactivity

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