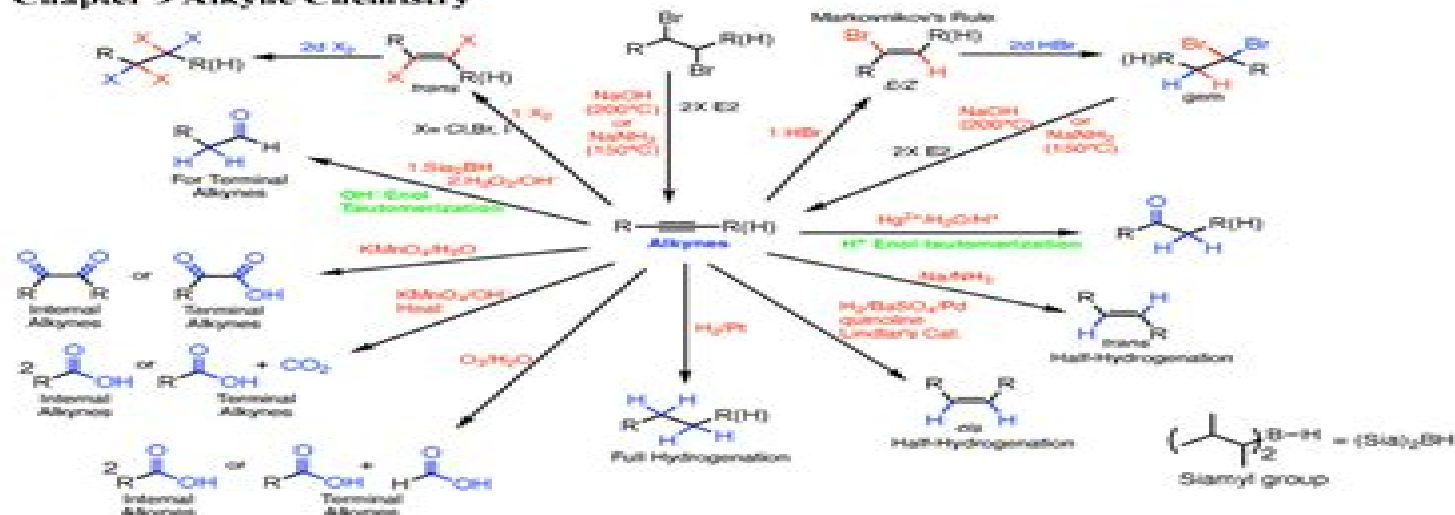
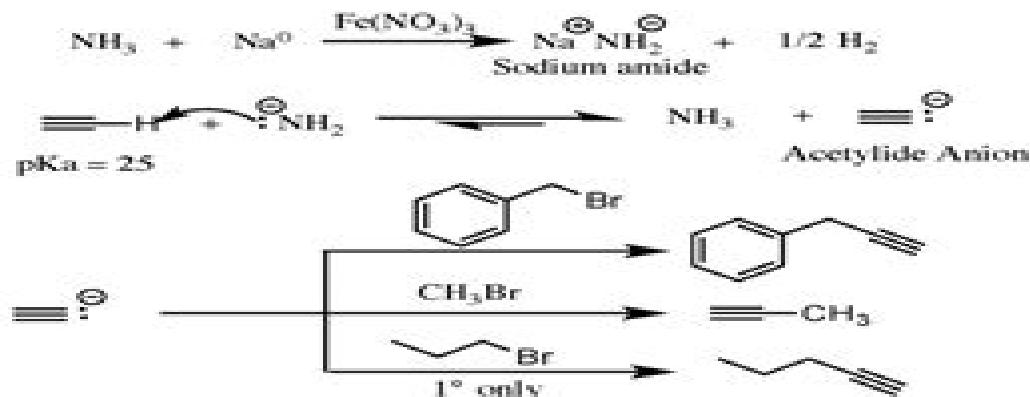


Chapter 9 Alkyne Chemistry



1. Acetylide Anions: Terminal triple bonds (alkynes, "yne" ending) have can be deprotonated with very strong bases. The most common base used is NaNH_2 , sodium amide, the conjugate base of a very weak acid, NH_3 ($\text{pK}_a = 35$). This makes the acetylide anion that is a great nucleophile for 1° alkyl halides, as well as methyl and benzyl halides. 2° and 3° undergo quick E2 elimination with acetylide anions because they are strong bases.



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