

# Chapter 10 Chemical Quantities Review

## Part 1: Vocabulary

Label the following descriptions with the correct vocabulary word.

- |                         |                                                           |
|-------------------------|-----------------------------------------------------------|
| _____ mass              | _____ percent composition                                 |
| _____ empirical formula | _____ empirical formula                                   |
| _____ molar mass        | _____ standard temperature and pressure (STP, 1 atm, 0°C) |
10. \_\_\_\_\_ is a description of the relative amounts of each element in a compound.
11. \_\_\_\_\_ composition
12. The lowest whole-number ratio of the atoms of the elements in a compound is the \_\_\_\_\_ formula.
13. The lowest structural formula and mass of any gas at STP is \_\_\_\_\_.
14. The lowest structural formula of a compound is \_\_\_\_\_.
15. The lowest structural formula of a compound is \_\_\_\_\_.
16. The lowest structural formula of a compound is \_\_\_\_\_.
17. The lowest structural formula of a compound is \_\_\_\_\_.
18. The lowest structural formula of a compound is \_\_\_\_\_.
19. The lowest structural formula of a compound is \_\_\_\_\_.
20. The lowest structural formula of a compound is \_\_\_\_\_.

## Part 2: Section Concept Review

### 10.1 The Mole: A Measurement

Complete the following section by using your knowledge of the terms and your understanding of the concepts involved with the section of the textbook. Complete the following section by using your knowledge of the terms and your understanding of the concepts involved with the section of the textbook.

1. The mass of a substance is a measurement of the mass of a substance. The mass of a substance is a measurement of the mass of a substance. The mass of a substance is a measurement of the mass of a substance.
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#### Section 10.1

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#### Section 10.2

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Rinehart and Winston Staff**



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Study Guide to Accompany Basics for Chemistry Martha Mackin,2012-12-02 Study Guide to Accompany Basics for Chemistry is an 18 chapter text designed to be used with Basics for Chemistry textbook Each chapter contains Overview Topical Outline Skills and Common Mistakes which are all keyed to the textbook for easy cross reference The Overview section summarizes the content of the chapter and includes a comprehensive listing of terms a summary of general concepts and a list of numerical exercises while the Topical Outline provides the subtopic heads that carry the corresponding chapter and section numbers as they appear in the textbook The Fill in Multiple Choice are two sets of questions that include every concept and numerical exercise introduced in the chapter and the Skills section provides developed exercises to apply the new concepts in the chapter to particular examples The Common Mistakes section is designed to help avoid some of the errors that students make in their effort to learn chemistry while the Practical Test section includes matching and multiple choice questions that comprehensively cover almost every concept and numerical problem in the chapter After briefly dealing with an overview of chemistry this book goes on exploring the concept of matter energy measurement problem solving atom periodic table and chemical bonding These topics are followed by discussions on writing names and formulas of compounds chemical formulas and the mole chemical reactions calculations based on equations gases and the properties of a liquid The remaining chapters examine the

solutions acids bases salts oxidation reduction reactions electrochemistry chemical kinetics and equilibrium and nuclear organic and biological chemistry This study guide will be of great value to chemistry teachers and students

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Chemical Engineering Progress ,1991

*Holt Chemistry* Ralph Thomas Myers,2004

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