

Chapter 10: Chemical Quantities

Section 1 Notes – The Mole: A Measurement of Matter (Read pages 287-296)

Mole

- chemist _____ for number of _____
- equals _____ number = _____

Representative Particles

- atoms – _____
- molecules – _____
- formula units – _____
- ions – _____

Conversions

- _____ mole equals
 - _____ atoms, molecules, formula units, ions, things

Practice Problems (Do Together)

3. How many moles is 2.80×10^{24} atoms of Si?
4. How many moles if 2.17×10^{23} representative particles of bromine?
5. How many atoms are in 1.14 mol SO_2 ?
6. How many moles are in 4.65×10^{24} molecules of NO_2 ?

Additional Practice Problems

1. How many moles are equal to 9.03×10^{24} atoms of mercury?
2. How many atoms are in 1.00 mole of sucrose, $\text{C}_{12}\text{H}_{22}\text{O}_{11}$?
3. How many atoms of O are in 3.65 moles of $\text{C}_{12}\text{H}_{22}\text{O}_{11}$?

Chapter 10 Chemical Quantities Laboratory Answers

Theodore Lawrence Brown



Chapter 10 Chemical Quantities Laboratory Answers:

Basic Chemistry Steven S Zumdahl,1996 **Purification of Laboratory Chemicals** W.L.F. Armarego,2022-08-27

Purification of Laboratory Chemicals Part One Physical Techniques Chemical Techniques Organic Chemicals Ninth Edition describes contemporary methods for the purification of chemical compounds The work includes tabulated methods taken from literature for purifying thousands of individual commercially available chemical substances To help in applying this information the more common processes currently used for purification in chemical laboratories and new methods are discussed For dealing with substances not separately listed another chapter is included setting out the usual methods for purifying specific classes of compounds Laboratory workers whether carrying out research or routine work will invariably need to consult this book Apart from the procedures described the large amount of physical data about listed chemicals is essential This fully updated revised and expanded new edition includes the purification of many new substances that have been available commercially since 2017 along with previously available substances which have found new applications Features empirical formulae and formula weights for every entry References all important applications of each substance Includes updated CAS registry numbers Covers the latest commercial chemical products including pharmaceutical chemicals and safety hazard materials Provides expanded coverage of laboratory work practices and purification methods

Chemistry Hall,1995-12

Fundamentals of Sustainable Chemical Science Stanley E. Manahan,2009-03-10 Written

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