

**6. What is meant by atomicity? Name a diatomic element.**

**Solution:**

A set of atoms of the same type together forms a molecule of the element. The number of atoms in a molecule of an element is called its atomicity.

Examples of diatomic elements are  $H_2$  – Hydrogen,  $O_2$  – Oxygen,  $N_2$  – Nitrogen

**7. a) Explain the terms valency and variable valency**

**b) How are the elements with variable valency named? Explain with an example.**

**Solution:**

a) Valency is the capacity of an atom to lose, gain or share atoms during a chemical reaction is called its valency. Sometimes atom of an element can lose more electron than they are present which means they lose electron from its penultimate shell. Such an element is said to exhibit variable valency.

b) If an element exhibits two different positive valencies, then

1. For the lower valency, use the suffix -OUS at the end of the name of the metal
2. For the higher valency, use the suffix -IC at the end of the name of the metal.

**8. Give the formula and valency of**

- a) aluminate
- b) chromate
- c) aluminium
- d) cupric

**Solution:**

Name	Formula	Valency
Aluminate	$AlO_2$	-2
Chromate	$CrO_4$	-2
Aluminium	Al	+3
Cupric	Cu	+2

**9. a) What is a chemical formula?**

**b) What is the significance of a formula? Give an illustrate.**

**Solution:**

a) Chemical formula is a symbolic representation of the number of atoms present in a molecule of that substance.

b) Significance of Chemical Formula

Chemical formula is very important in finding information about chemical compounds as it tells us about the elements and the number of atoms in a substance

Example - Salt -  $NaCl$ , ethanol  $C_2H_6O$  because the molecule of ethanol contains two Carbon, 6 Hydrogen and 1 Oxygen atom.

# Chapter 1 Introduction To Chemistry Worksheet Answers

**Rodney J. Sime**



## **Chapter 1 Introduction To Chemistry Worksheet Answers:**

**Chemistry (Teacher Guide)** Dr. Dennis Englin, 2018-02-26 This book was created to help teachers as they instruct students through the Master's Class Chemistry course by Master Books. The teacher is one who guides students through the subject matter, helps each student stay on schedule and be organized, and is their source of accountability along the way. With that in mind, this guide provides additional help through the laboratory exercises as well as lessons, quizzes, and examinations that are provided along with the answers. The lessons in this study emphasize working through procedures and problem solving by learning patterns. The vocabulary is kept at the essential level. Practice exercises are given with their answers so that the patterns can be used in problem solving. These lessons and laboratory exercises are the result of over 30 years of teaching home school high school students and then working with them as they proceed through college. Guided labs are provided to enhance instruction of weekly lessons. There are many principles and truths given to us in Scripture by the God that created the universe and all of the laws by which it functions. It is important to see the hand of God and His principles and wisdom as it plays out in chemistry. This course integrates what God has told us in the context of this study.

**Features** Each suggested weekly schedule has five easy-to-manage lessons that combine reading and worksheets. Worksheets, quizzes, and tests are perforated, and three-hole punched materials are easy to tear out, hand out, grade, and store. Adjust the schedule and materials needed to best work within your educational program. Space is given for assignments, dates. There is flexibility in scheduling. Adapt the days to your school schedule.

**Workflow** Students will read the pages in their book and then complete each section of the teacher guide. They should be encouraged to complete as many of the activities and projects as possible as well. Tests are given at regular intervals with space to record each grade.

**About the Author** DR DENNIS ENGLIN earned his bachelor's from Westmont College, his master of science from California State University, and his EdD from the University of Southern California. He enjoys teaching animal biology, vertebrate biology, wildlife biology, organismic biology, and astronomy at The Master's University. His professional memberships include the Creation Research Society, the American Fisheries Association, Southern California Academy of Sciences, Yellowstone Association, and Au Sable Institute of Environmental Studies.

**The Art of Teaching Science** Jack Hassard, Michael Dias, 2013-07-04 The Art of Teaching Science emphasizes a humanistic, experiential, and constructivist approach to teaching and learning and integrates a wide variety of pedagogical tools. Becoming a science teacher is a creative process, and this innovative textbook encourages students to construct ideas about science teaching through their interactions with peers, mentors, and instructors, and through hands-on, minds-on activities designed to foster a collaborative, thoughtful learning environment. This second edition retains key features such as inquiry-based activities and case studies throughout while simultaneously adding new material on the impact of standardized testing on inquiry-based science and explicit links to science teaching standards. Also included are expanded resources like a comprehensive website, a streamlined format, and updated content, making the experiential tools in the book even more useful.

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**Physical Chemistry Calculations** Rodney J. Sime, 2005 Physical Chemistry Calculations is a practical guide for students and instructors who want to learn how to use the most popular spreadsheet and computational software to solve problems in physical chemistry The book provides students with a complementary approach to the chemistry and physics they are learning in the classroom Physical Chemistry Calculations

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