#### WORKSHEET - ASSIGNING OXIDATION NUMBERS

Name Period

#### Oxidation Number Rules

1. A pure element has an oxidation number of 0.

The oxidation number of an element in a monatomic ion equals the charge of the ion.

3. The sum of the oxidation number of all the elements in a compound equals 0.

4. The sum of the oxidation numbers of all the elements in a polyatomic ion equals the charge on the ion.

Directions In the following questions, give the oxidation number of the indicated atoms/ion

N in N<sub>2</sub>O<sub>3</sub>
 N \*\*\*

2. S in H<sub>2</sub>SO<sub>4</sub> S<sup>4\*</sup>

3. C C\*

4 Cin CO C"

5 Nain NaCl Na\*

6 H in H<sub>2</sub>O H\*

7. Bain BaCl<sub>2</sub> Ba<sup>1+</sup>

8. N in NO<sub>2</sub> N<sup>3\*</sup>

S in Al<sub>3</sub>S<sub>3</sub>
 S<sup>3</sup>

10 S in HSO. 5\*\*

11. Cl in Fe(ClO<sub>2</sub>)<sub>3</sub> Cl<sup>3\*</sup>

12. Fe in Fe(ClO<sub>2</sub>)<sub>3</sub> Fe<sup>3\*</sup>

13. N in NO<sub>3</sub> No.

14 Cu2\* Cu2\*

15. Zn2\* Zn2\*

16 Cin CH<sub>4</sub> C\*

17. Mn in MnO<sub>2</sub> Mn\*\*

13 S in SO<sub>2</sub>3 S\*\*

19 Mg2 Mg2

20 CT CT

21 Oz Oz"

22 P. P.

23 Na in NaoS Na\*

24. S in H<sub>2</sub>S S<sup>2</sup>

25. Ca2+ Ca2+

26 C in CN C\*

27. H in OH H"

28. Mn in KMnO<sub>4</sub> Mn<sup>5</sup>

29. I in Mg(IO<sub>3</sub>)<sub>2</sub> I<sup>50</sup>

30. C in C<sub>2</sub>O<sub>4</sub>3- C3-

# **Cavalcade Oxidation State Worksheet Answer Key**

**WJ Hussar** 

**Cavalcade Oxidation State Worksheet Answer Key:** 

Decoding Cavalcade Oxidation State Worksheet Answer Key: Revealing the Captivating Potential of Verbal Expression

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