Name Date	Class

CHAPTER 8 STUDY GUIDE

Covalent Bonding

Section 8.1 The Covalent Bond

In your textbook, read about the nature of covalent bonds.

Use each of the terms below just once to complete the passage.

covalent bond	molecule	sigma bond	exothermic	pi bond	
When sharing of el-	ectrons occurs, the attac	hment between atom	s that results is called		
	. When suc				
released, and the proce	ess is (2)	. When t	two or more atoms bon-	d by means of	
electron sharing, the resulting particle is called a(n) (3)					
are centered between the two atoms, the attachment is called a(n) (4		4)	If the		
sharing involves the or	verlap of parallel orbital	s, the attachment is c	alled a(n) (5)		
6. In what form do e	e choice that best comp dements such as hydrog	en, nitrogen, and oxy	gen normally occur?		
	a. as single atoms		 as molecules containing three atoms 		
	containing two atoms		lecules containing four	atoms	
	ons are shared in a doub			encorporate	
a. none	b. one	e, two	d. I	our	
a. two molecules	e distance between s of the same substance, in two attached atoms.		 c. the nuclei of two attached atoms. d. the orbitals of two attached atoms. 		
9. Which of the follo	owing relationships rela-	ting to bond length is	generally correct?		
a. the shorter the	bond, the stronger the l	bond			
b. the shorter the	bond, the weaker the b	ond			
c. the shorter the	bond, the fewer the ele	ectrons in it			
d. the shorter the	bond, the lower the bor	nd dissociation energ	DY.		

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K Morrison

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