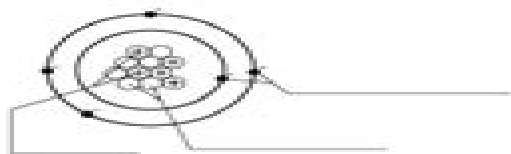


Basic Atomic Structure

1. Label the parts of an atom in the diagram below



- What type of charge does a proton have? _____
- What type of charge does a neutron have? _____
- What type of charge does an electron have? _____
- Which two subatomic particles are located in the nucleus of an atom? _____

2. An element is represented by its chemical symbol along with a few numbers. Based on the example on the right, fill in the numbers of protons (P), neutrons (N), and electrons (E) for the following elements.

| | |
|--------|-----------------------|
| 8 | ← Atomic Number |
| O | ← Symbol |
| Oxygen | ← Name |
| 15.999 | ← Average atomic mass |

| | |
|--------|----------|
| 6 | #P _____ |
| C | #N _____ |
| Carbon | #E _____ |
| 12.011 | |

| | |
|--------|----------|
| 10 | #P _____ |
| Ne | #N _____ |
| Neon | #E _____ |
| 20.108 | |

| | |
|-----------|----------|
| 19 | #P _____ |
| K | #N _____ |
| Potassium | #E _____ |
| 39.098 | |

3. Complete the table using your knowledge of the periodic table.

| Symbol | Atomic Number | Mass Number | Number of Protons | Number of Electrons | Number of Neutrons |
|--------|---------------|-------------|-------------------|---------------------|--------------------|
| Na | | | 11 | | 12 |
| K | | 39 | | 19 | |
| | | | 38 | | 50 |
| F | | | | 9 | 10 |
| | 20 | 40 | | 20 | |
| | 50 | | | 50 | 69 |
| I | 53 | 127 | | | |

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