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<p><b>Hydrogen Bonding:</b></p> <p>1. <b>Molecular Oxygen, <math>H_2O</math></b>  <b>Example:</b> Ice, Sea, Rain</p> <p>2. <b>Ammonia, <math>NH_3</math></b>  <b>Example:</b> Rain, Sea, Rain</p> <p>3. <b>Alcohols, <math>C_nH_{2n+2}O</math></b>  <b>Example:</b> Rain, Sea, Rain</p> <p>4. <b>Carbohydrates, <math>C_nH_{2n}O_n</math></b>  <b>Example:</b> Rain, Sea, Rain</p> <p>5. <b>Hydrogen Bonding in Water</b>  <b>Example:</b> Rain, Sea, Rain</p> <p>6. <b>Hydrogen Bonding in Ice</b>  <b>Example:</b> Rain, Sea, Rain</p>	<p><b>Hydrogen Bonding in Water:</b> Water molecules are held together by hydrogen bonds, which are responsible for the high boiling point of water.</p> <p><b>Hydrogen Bonding in Ice:</b> Ice is a solid at room temperature because of the hydrogen bonds between water molecules.</p> <p><b>Hydrogen Bonding in Ammonia:</b> Ammonia molecules are held together by hydrogen bonds, which are responsible for the high boiling point of ammonia.</p> <p><b>Hydrogen Bonding in Alcohols:</b> Alcohols are held together by hydrogen bonds, which are responsible for their high boiling points.</p> <p><b>Hydrogen Bonding in Carbohydrates:</b> Carbohydrates are held together by hydrogen bonds, which are responsible for their high boiling points.</p>
<p>7. <b>Hydrogen Bonding in DNA</b>  <b>Example:</b> DNA</p> <p>8. <b>Hydrogen Bonding in Proteins</b>  <b>Example:</b> Proteins</p>	<p><b>Hydrogen Bonding in DNA:</b> DNA is a double helix structure held together by hydrogen bonds between the base pairs.</p> <p><b>Hydrogen Bonding in Proteins:</b> Proteins are held together by hydrogen bonds, which are responsible for their specific three-dimensional shapes.</p>

<sup>2</sup> <http://www.fishbase.org>

**Dissemination:** *Journal of Interpersonal Violence*, 2006, 21(12), 1599-1610. doi:10.1177/0886260506288606

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# Chang 8th Edition Answers

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