

## Chapter 10 - Chemical Quantities

### 10.1 The Mole: A Measurement of Matter:

#### Mole:

- A counting unit
- $6.02 \times 10^{23}$  representative particles (atom, molecule, formula unit, ion) of a substance = **Avogadro's Number**

Converting # of particles to Moles:

$$\text{Moles} = \cancel{\text{Representative particles}} \left( \frac{1 \text{ mole}}{6.02 \times 10^{23} \cancel{\text{representative particles}}} \right)$$

Converting Moles to # of particles:

$$\text{Representative particles} = \cancel{\text{moles}} \left( \frac{6.02 \times 10^{23} \cancel{\text{representative particles}}}{1 \cancel{\text{mole}}} \right)$$

#### Molar Mass:

- Molar mass of an element = atomic mass (g)  
Ex. 24.3 g = molar mass of Mg
- Molar mass = g/mol
- Molar mass of any element contains 1 mol or  $6.02 \times 10^{23}$  atoms of that element
- Molar mass of a compound = sum of the atomic masses of each element contained in 1 mole of the compound.

### 10.2 Mole-Mass and Mole-Volume Relationships:

#### Mole-Mass Relationship:

- Use molar mass (g/mol) of an element or compound to convert b/w the mass of a substance and the moles of a substance

$$\text{Mass (grams)} = \cancel{\# \text{ of moles}} \left( \frac{\cancel{\text{mass (grams)}}}{1 \cancel{\text{mole}}} \right)$$

$$\cancel{\# \text{ of Moles}} = \cancel{\text{Mass (grams)}} \left( \frac{1 \cancel{\text{mole}}}{\cancel{\text{mass (grams)}}} \right)$$

# Chapter 10 Chemical Quantities Packet Answers

**Harold Maurice Frost**



## Chapter 10 Chemical Quantities Packet Answers:

**Human Anatomy & Physiology Coloring Workbook** Anderson, 2008-04-25 This valuable student resource is intended for use in the undergraduate human anatomy and physiology class The latest edition of Human Anatomy and Physiology Coloring Workbook is designed to help students learn introductory anatomy and physiology and is organized to complement the leading texts in the field Virtually every structure of the human body typically studied in an introductory course is examined Chapters are short concise and complete enabling the student to master smaller sections of information in a cohesive manner 2009 292 pages

**Prentice Hall Physical Science Concepts in Action Program Planner National Chemistry Physics Earth Science**, 2003-11 Prentice Hall Physical Science Concepts in Action helps students make the important connection between the science they read and what they experience every day Relevant content lively explorations and a wealth of hands on activities take students understanding of science beyond the page and into the world around them Now includes even more technology tools and activities to support differentiated instruction

**Fundamentals of Fire Fighter Skills** International Association of Fire Chiefs, National Fire Protection Association, 2008-08-07 This resource is designed to encourage critical thinking and aid comprehension of the course material through use of the following materials Case studies and corresponding questions Figure labeling exercises Crossword puzzles Matching fill in the blank short answer and multiple choice questions Skill Drill activities Answer key with page references

*Human Anatomy and Physiology Coloring Workbook and Study Guide* Paul D. Anderson, 1990

**Workbook for Organic Chemistry** Jerry Jenkins, 2009-12-25 With authors who are both accomplished researchers and educators Vollhardt and Schore's Organic Chemistry is proven effective for making contemporary organic chemistry accessible introducing cutting edge research in a fresh student friendly way A wealth of unique study tools help students organize and understand the substantial information presented in this course And in the sixth edition the themes of understanding reactivity mechanisms and synthetic analysis to apply chemical concepts to realistic situations has been strengthened New applications of organic chemistry in the life sciences industrial practices green chemistry and environmental monitoring and clean up are incorporated This edition includes more than 100 new or substantially revised problems including new problems on synthesis and green chemistry and new challenging problems

**Hazardous Materials Awareness and Operations, Student Workbook** Iafc, 2009-10-02 This Student Workbook contains exercises to reinforce what you will learn in both the Hazardous Materials Awareness and Operations textbook and in the classroom The exercises cover the 2008 edition of NFPA 472 Standard for Competence of Responders to Hazardous Materials Weapons of Mass Destruction Incidents and are designed to encourage critical thinking and aid comprehension through a variety of exercises for each chapter including Multiple Choice Numerous multiple choice questions prepare you for exams Fire Alarms Scenarios help you develop your decision making skills by describing emergencies and challenging you to come up with solutions Skill Drills Test your skills with photo jumbles and caption fill ins

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