

# Chemistry

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(Chapter 10) (Haloalkanes and Haloarenes) (Exercises)

(Class 12)



## Question 10.15:

Write the mechanism of the following reaction:

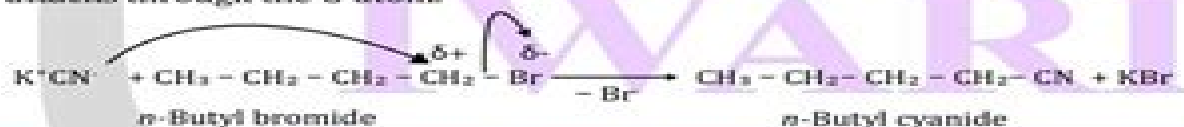


## Answer 10.15:

The given reaction is:



The given reaction is an  $\text{S}_{\text{N}}2$  reaction. In this reaction,  $\text{CN}^-$  acts as the nucleophile and attacks the carbon atom to which Br is attached.  $\text{CN}^-$  ion is an ambident nucleophile and can attack through both C and N. In this case, it attacks through the C-atom.



## Question 10.16:

Arrange the compounds of each set in order of reactivity towards  $\text{S}_{\text{N}}2$  displacement:

(i) 2-Bromo-2-methylbutane, 1-Bromopentane, 2-Bromopentane

(ii) 1-Bromo-3-methylbutane, 2-Bromo-2-methylbutane, 2-Bromo-3-methylbutane

(iii) 1-Bromobutane, 1-Bromo-2, 2-dimethylpropane, 1-Bromo-2-methylbutane, 1-Bromo-3-Methylbutane.

# Chapter 10 Self Check Answers For Chemistry

**James M. Dorn, Barbara M. Hopkins**



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