# Identifying Reaction Types & Balancing Equations

Step 1: Name the type of chemical reaction	Step 2: Balance the following chemical equations:
1. Ex: Synthesis	N₂ + H₂ → NH₃
2.	_ KCIO <sub>3</sub> → _ KCI + _O₂
3.	NaCl + F₂ → NaF + Cl₂
4.	H₂ + O₂ →H₂O
5.	AgNO <sub>3</sub> + MgCl <sub>2</sub> → AgCl + Mg(NO <sub>3</sub> ) <sub>2</sub>
6.	AIBr3 *K2SO4 →KBr *AI2(SO4)3
7.	CH4 +O2 →CO2 +H2O
8.	C3H8 +O2 →CO2 +H2O
9.	FeCl <sub>3</sub> *NaOH →Fe(OH) <sub>3</sub> *NaCl
10.	P + O₂ → P₂O₅
11.	Na + HzO → NaOH + Hz
12.	Ag <sub>2</sub> O → Ag + O <sub>2</sub>
13.	CO2 + H2O → C6H12O6 + O2
14,	K +MgBr₂ → KBr +Mg
15.	HNO <sub>3</sub> (aq) + Ba(OH) <sub>2</sub> (aq) → Ba(NO <sub>3</sub> ) <sub>2</sub> (aq) + H <sub>2</sub> O (I)
16.	$C_5H_{12}(g) + C_2(g) \rightarrow C_2(g) + H_2O(g)$
17.	Al(s) + Fe <sub>2</sub> O <sub>3</sub> (aq) → Al <sub>2</sub> O <sub>3</sub> (aq) + Fe(s)
18.	Al(s) + O₂(g) → Al₂O₃ (s)

John Olmsted, Gregory M. Williams

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