Name: Date:

#### **Balancing Chemical Equations**

Balance the following chemical equations.

1. \_\_\_ Fe + \_\_\_ 
$$H_2SO_4$$
  $\longrightarrow$  \_\_\_  $Fe_2(SO_4)_3 + ___  $H_2$$ 

3. \_\_\_ 
$$SiCl_4(\ell)$$
 + \_\_\_  $H_2O(\ell)$   $\longrightarrow$  \_\_\_  $SiO_2(s)$  + \_\_\_  $HCl(aq)$ 

5. \_\_\_ 
$$NH_3$$
 + \_\_\_  $O_2$   $\longrightarrow$  \_\_\_  $NO$  + \_\_\_  $H_2O$ 

7. \_\_\_ 
$$SiO_2$$
 + \_\_\_  $HF$   $\longrightarrow$  \_\_\_  $SiF_4$  + \_\_\_  $H_2O$ 

9. \_\_\_\_ (NH<sub>4</sub>)<sub>3</sub>PO<sub>4</sub> + \_\_\_\_ Pb(NO<sub>3</sub>)<sub>4</sub> 
$$\longrightarrow$$
 \_\_\_\_ Pb<sub>3</sub>(PO<sub>4</sub>)<sub>4</sub> + \_\_\_\_ NH<sub>4</sub>NO<sub>3</sub>

# **Balancing Chemical Equations Worksheet 50**

Holt Rinehart & Winston, Holt, Rinehart and Winston Staff

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