Balancing Equations Worksheet - Answers

Note to students: It is acceptable to leave spaces blank when balancing equations – blank spaces are interpreted as containing the number "1".

- 1 Na₃PO₄ + 3 KOH → 3 NaOH + 1 K₃PO₄
- 2) 1 MgF₂ + 1 Li₂CO₃ → 1 MgCO₃ + 2 LiF
- 3) 1 P₄ + 3 O₂ → 2 P₂O₃
- 4) 2 RbNO₃ + 1 BeF₂ → 1 Be(NO₃)₂ + 2 RbF
- 5) 2 AgNO₃ + 1 Cu → 1 Cu(NO₃)₂ + 2 Ag
- 6) 1 CF₄ + 2 Br₂ → 1 CBr₄ + 2 F₂
- 7) 2 HCN + 1 CuSO₄ → 1 H₂SO₄ + 1 Cu(CN)₂
- 8) 1 GaF₃ + 3 Cs → 3 CsF + 1 Ga
- 9) 1 BaS + 1 PtF₂ → 1 BaF₂ + 1 PtS
- 10) 1 N₂ + 3 H₂ → 2 NH₃
- 11) 2 NaF + 1 Br₂ → 2 NaBr + 1 F₂
- 12) 1 Pb(OH)₂ + 2 HCl → 2 H₂O + 1 PbCl₂
- 13) 2 AlBr₃ + 3 K₂SO₄ → 6 KBr + 1 Al₂(SO₄)₃
- 14) 1 CH₄ + 2 O₂ → 1 CO₂ + 2 H₂O
- 15) 2 Na₃PO₄ + 3 CaCl₂ → 6 NaCl + 1 Ca₃(PO₄)₂
- 16) 2 K + 1 Cl₂ → 2 KCl
- 17) 2 Al + 6 HCl → 3 H₂ + 2 AlCl₃
- 18) 1 N₂ + 3 F₂ → 2 NF₃
- 19) 1 SO₂ + 2 Li₂Se → 1 SSe₂ + 2 Li₂O
- 20) 2 NH₃ + 1 H₂SO₄ → 1 (NH₄)₂SO₄

Balancing Equations Activity Sheet Answers

Lisa Willman

Balancing Equations Activity Sheet Answers:

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learning because they have the comfort of a more familiar general business context for thinking about the accounting issue at hand Work Pap/Sg Ch 1-20, C21 Acct Robert M. Swanson, Kenton E. Ross, 1991-03 Elementary Laboratory

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