

Activity 10

Classifying reactions

On the Go: 10 min

Types of chemical reactions and their characteristics



Introduction: Chemical reactions show how **compounds** and **elements** react with one another. An element is an substance consisting of only one of a kind. Such as the elements of hydrogen, oxygen and fluorine. A compound is a substance made of more than one kind of atoms. Such as water, which is made up of hydrogen.

Question: How are chemical reactions classified?

1. Study these chemical reactions and be classified as one of four types. Using the chemical equations in the boxes on the right, decide the following statements in the right or wrong.

- | | |
|---|------------------------------|
| <input checked="" type="checkbox"/> The reaction is broken down into two or more products. | a. Decomposition |
| <input checked="" type="checkbox"/> Aqueous is combined with oxygen to produce carbon dioxide and water. | a. Decomposition |
| <input checked="" type="checkbox"/> Two or more compounds combine to form one product. | c. Single replacement |
| <input checked="" type="checkbox"/> Two compounds react to form two different compounds. | b. Double replacement |
| <input checked="" type="checkbox"/> A compound reacts with an element to form a new compound and a different element. | d. Combustion |

2. Study these reactions and be classified as one of four types. Using the chemical equations in the boxes on the right, decide the reaction as a synthesis, decomposition, single replacement, or double replacement reaction.

- | | |
|--|---------------------------|
| a. <input checked="" type="checkbox"/> $2\text{H}_2\text{O} \rightarrow 2\text{H}_2 + \text{O}_2$ | Decomposition |
| b. <input checked="" type="checkbox"/> $2\text{H}_2 + \text{O}_2 \rightarrow 2\text{H}_2\text{O}$ | Synthesis |
| c. <input checked="" type="checkbox"/> $\text{Zn} + 2\text{HCl} \rightarrow \text{ZnCl}_2 + \text{H}_2$ | Single replacement |
| d. <input checked="" type="checkbox"/> $\text{NaOH} + \text{HCl} \rightarrow \text{NaCl} + \text{H}_2\text{O}$ | Double replacement |
| e. <input checked="" type="checkbox"/> $\text{Zn} + 2\text{HCl} \rightarrow \text{ZnCl}_2 + \text{H}_2$ | Single replacement |

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Siddappa N.Byrareddy

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the author's name.

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