

BALANCING CHEMICAL EQUATIONS

Name _____

Rewrite and balance the equations below.

1. $\text{N}_2 + \text{H}_2 \rightarrow \text{NH}_3$ _____
2. $\text{KClO}_3 \rightarrow \text{KCl} + \text{O}_2$ _____
3. $\text{NaCl} + \text{F}_2 \rightarrow \text{NaF} + \text{Cl}_2$ _____
4. $\text{H}_2 + \text{O}_2 \rightarrow \text{H}_2\text{O}$ _____
5. $\text{AgNO}_3 + \text{MgCl}_2 \rightarrow \text{AgCl} + \text{Mg}(\text{NO}_3)_2$ _____
6. $\text{AlBr}_3 + \text{K}_2\text{SO}_4 \rightarrow \text{KBr} + \text{Al}_2(\text{SO}_4)_3$ _____
7. $\text{CH}_4 + \text{O}_2 \rightarrow \text{CO}_2 + \text{H}_2\text{O}$ _____
8. $\text{C}_3\text{H}_8 + \text{O}_2 \rightarrow \text{CO}_2 + \text{H}_2\text{O}$ _____
9. $\text{C}_8\text{H}_{18} + \text{O}_2 \rightarrow \text{CO}_2 + \text{H}_2\text{O}$ _____
10. $\text{FeCl}_3 + \text{NaOH} \rightarrow \text{Fe}(\text{OH})_3 + \text{NaCl}$ _____
11. $\text{P} + \text{O}_2 \rightarrow \text{P}_2\text{O}_5$ _____
12. $\text{Na} + \text{H}_2\text{O} \rightarrow \text{NaOH} + \text{H}_2$ _____
13. $\text{Ag}_2\text{O} \rightarrow \text{Ag} + \text{O}_2$ _____
14. $\text{S}_8 + \text{O}_2 \rightarrow \text{SO}_3$ _____
15. $\text{CO}_2 + \text{H}_2\text{O} \rightarrow \text{C}_6\text{H}_{12}\text{O}_6 + \text{O}_2$ _____
16. $\text{K} + \text{MgBr}_2 \rightarrow \text{KBr} + \text{Mg}$ _____
17. $\text{HCl} + \text{CaCO}_3 \rightarrow \text{CaCl}_2 + \text{H}_2\text{O} + \text{CO}_2$ _____

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If8767

H Kauffman



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