Name:

Writing Balanced Chemical Equations

Write balanced chemical equations for the following reactions:

- aluminum bromide + chlorine → aluminum chloride + bromine
- (b) calcium sulfite → calcium oxide + sulfur dioxide
- © phosphoric acid + calcium hydroxide → calcium phosphate + water
- d silver nitrate + ammonium carbonate → silver carbonate + ammonium nitrate
- (e) lithium hydrogen carbonate → lithium carbonate + water + carbon dioxide
- f magnesium + silver acetate → magnesium acetate + silver
- (g) aluminum bromide + potassium sulfate → potassium bromide + aluminum sulfate
- (h) sodium chloride + fluorine -- sodium fluoride + chlorine
- i) copper + sulfuric acid → copper sulfate + water + oxygen
- (j) zinc + iron (III) chloride → zinc chloride + iron

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Frank Schaffer Publications, Joan DiStasio

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Basic Chemistry G. William Daub, William S. Seese, 1996 Written in a style and language that users without science backgrounds can understand This best selling introduction to the basic principles of chemistry draws on the reader s own experiences through analogies and cartoons to learn difficult concepts The clear systematic thinking approach to problem solving has also been highly praised by reviewers and users alike Countdown sections in each chapter consisting of five review questions keyed to previous material provide readers with a basis for material introduced in the new chapter Study exercises found immediately after new topics are introduced reinforce chapter problem material You and Chemistry marginal application icon relates chemistry to the real world End of chapter essays entitled Elements and Compounds relate the applications of specific elements or compounds to the readers life George Facer's Edexcel A Level Chemistry Student Book 1 George Facer, 2015-05-15 Exam Board Edexcel Level AS A level Subject Chemistry First Teaching September 2015 First Exam June 2016 Help higher achieving students to maximise their potential with a focus on independent learning assessment advice and model assessment answers in this new edition of George Facer's best selling textbook Encourages independent learning with notes and clear explanations throughout the content Strengthens understanding with worked examples of chemical equations and calculations Stretches the students with a bank of questions at the end of each chapter Teacher Instruction and Student Performance in Balancing Provides assessment guidance and sample answers **Chemical Equations** William Levi Yarroch, 1983 Basic Chemistry William S. Seese, Guido H. Daub, 1977 Cambridge IGCSETM Chemistry Student's Book (Collins Cambridge IGCSETM) Chris Sunley, 2021-04-22 Collins IGCSE Chemistry provides complete coverage of the latest Cambridge IGCSE syllabus for Chemistry and is packed full of questions in depth content practical investigative skills features and more Foundations of College Chemistry Morris Hein, Susan Arena, 1993 This is the Student Study Guide to accompany Hein's Foundations of College Chemistry Alternate Edition 12th Edition Lower Secondary Science Workbook: Stage 8 (Collins Cambridge Lower Secondary Science) Collins, 2022-02-03 Inspire and engage your students with this Lower Secondary Science course from Collins offering comprehensive coverage of the new curriculum framework including suggested practical investigations and Thinking and Working Scientifically skills Journal of Chemical Education, 1925 Includes Report of New England Association of Chemistry Teachers and Proceedings of the Pacific Southwest Association of Chemistry Teachers General Chemistry William A. Nevill, 1967 Chemistry Homework Frank Schaffer Publications, Joan DiStasio, 1996-03 Includes the periodic table writing formulas balancing equations stoichiometry problems and more **General Chemistry** Darrell D. Student Study Guide to accompany Foundations of College Chemistry, 11th Ebbing, Larry K. Krannich, Joan I. Senyk, 1996 Edition Morris Hein, Susan Arena, 2003-09-03 Work more effectively and gauge your progress along the way This Student Study Guide that accompanies Hein's Foundations of College Chemistry 11th Edition contains a review and self evaluation

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