

## Molecular Models Lab Covalent Compounds

### Introduction

The purpose of this lab is to evaluate the geometries and polarities of simple covalent molecules. Many physical and chemical properties of compounds depend upon the shape or geometry of the molecule and its polarity. A hemoglobin molecule, for example, becomes ineffective as an oxygen transport vehicle if its shape is altered. Molecular shape determines a compound's boiling point, freezing point, and viscosity. A molecule's polarity determines whether it dissolves in water or not.

### Molecular Geometries

The geometry of a small molecule can be predicted by examining the central atom. The valence electron pairs that surround the central atom will repel each other. The electron pairs will position themselves so that there is a maximum distance between them. This rule is called the **VSEPR** theory (Valence Shell Electron Pair Repulsion). The electron pairs include the covalent bonds (shared pairs) and the unshared pairs of electrons. The 5 simplest geometries and their characteristics are found in the following table:

No. of Electron Pairs on Central Atom	Geometry Name	Central Atom Characteristics	Comments
No central atom	Linear	$X - X$	If a molecule only contains 2 atoms it is linear.
2 shared, 0 unshared	Linear	$— X —$ or $= X =$	Double and triple bonds only count as a shared pair for geometry purposes.
3 shared, 0 unshared	Trigonal planar	$\begin{array}{c}   \\ — X — \end{array}$ or $\begin{array}{c}   \\ = X = \end{array}$	Bond angles are $120^\circ$
4 shared, 0 unshared	Tetrahedral	$\begin{array}{c}   \\ — X — \\   \end{array}$	Since the molecule is 3-dimensional, the bond angles are $109^\circ$ , not $90^\circ$ .
2 shared, 2 unshared	Bent	$\begin{array}{c}   \\ : X : \\   \end{array}$	The bond angle is slightly $< 109^\circ$ , not $90^\circ$ .
3 shared, 1 unshared	Pyramidal	$\begin{array}{c}   \\ — X : \\   \end{array}$	The bond angle is slightly $< 109^\circ$ , not $90^\circ$ .

# Conclusion For Molecular Models Lab In Chemistry

**Yi-Tong Ma**



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DR DENNIS ENGLIN earned his bachelor's from Westmont College, his master of science from California State University, and his EdD from the University of Southern California. He enjoys teaching animal biology, vertebrate biology, wildlife biology, organismic biology, and astronomy at The Master's University. His professional memberships include the Creation Research Society, the American Fisheries Association, Southern California Academy of Sciences, Yellowstone Association, and the Sable Institute of Environmental Studies.

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Computational Modelling and Simulations for Designing of Corrosion Inhibitors Dakeshwar Kumar Verma, Chandrabhan Verma, Jeenat Aslam, 2023-04-19 Computational Modeling and

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