

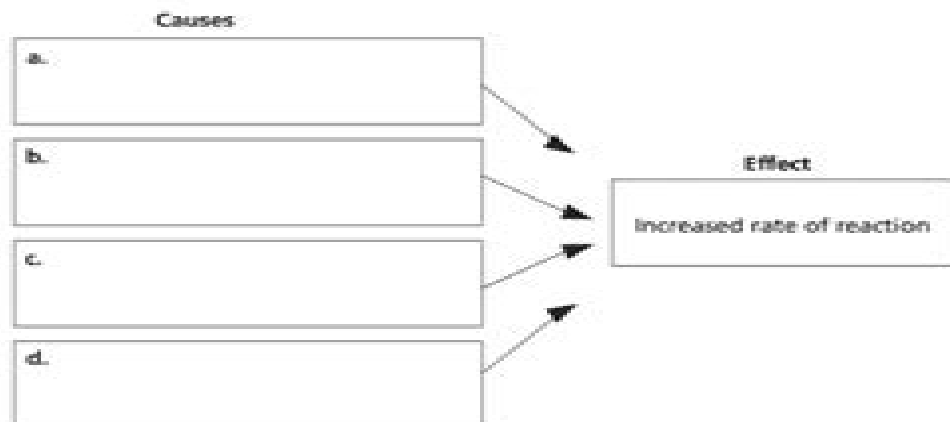
Chemical Reactions • Guided Reading and Study

## Controlling Chemical Reactions

*This section explains how energy is related to chemical reactions. It also describes how the rates of chemical reactions can be controlled.*

### Use Target Reading Skills

*As you read, identify the factors that can cause the rate of a chemical reaction to increase. Write the information in the graphic organizer below.*



### Energy and Reactions

1. The \_\_\_\_\_ is the minimum amount of energy needed to start a chemical reaction.
2. Is the following sentence true or false? All chemical reactions need a certain amount of activation energy to get started.  
\_\_\_\_\_
3. In a reaction that makes water from hydrogen gas and oxygen gas, where does the activation energy come from?  
\_\_\_\_\_  
\_\_\_\_\_
4. A reaction that releases energy is called a(n) \_\_\_\_\_.
5. A reaction that absorbs energy is called a (n) \_\_\_\_\_.

# Controlling Chemical Reactions Guided Reading And Study

**J Dewey**



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