

Chemical Reactions

Section 9.1 Reactions and Equations

In your textbook, read about evidence of chemical reactions.

For each statement, write *yes* if evidence of a chemical reaction is present. Write *no* if there is no evidence of a chemical reaction.

- | | |
|---------------|---|
| <u>yes</u> | 1. A tomato smells rotten. |
| <u>no</u> | 2. A drinking glass breaks into smaller pieces. |
| <u>no</u> | 3. A piece of ice melts. |
| <u>yes</u> | 4. Drain cleaner is mixed with water and the solution becomes warm. |
| <u>yes</u> | 5. Candle wax burns. |
| <u>yes/no</u> | 6. Molten candle wax solidifies. |
| <u>yes</u> | 7. Green leaves turn yellow and red as the seasons change. |
| <u>yes</u> | 8. Baking powder produces a gas that makes a cake rise. |

In your textbook, read about how to represent chemical reactions and how to balance chemical equations.

Use the terms below to complete the passage. Each term may be used once, more than once, or not at all.

arrow	plus sign	(s)	(l)
reactant	product	(g)	(aq)

The fuel for the space shuttle is hydrogen, which burns in oxygen to produce water vapor and energy. In this chemical reaction, hydrogen is a(n) **(9)** reactant, oxygen is a(n) **(10)** reactant, and water vapor is a(n) **(11)** product. In a chemical equation for this reaction, a(n) **(12)** arrow is used to separate hydrogen and oxygen from water vapor and energy. A(n) **(13)** plus sign is used to separate the symbols for hydrogen and oxygen. A(n) **(14)** (g) symbol is used to tell the state of hydrogen in the reaction, a(n) **(15)** (g) symbol is used for the state of oxygen, and a(n) **(16)** (g) symbol is used for the state of water vapor.

Describing Chemical Reactions Study Guide Answer Key

Neil E. Schore



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Table of Contents Describing Chemical Reactions Study Guide Answer Key

1. Understanding the eBook Describing Chemical Reactions Study Guide Answer Key
 - The Rise of Digital Reading Describing Chemical Reactions Study Guide Answer Key
 - Advantages of eBooks Over Traditional Books
2. Identifying Describing Chemical Reactions Study Guide Answer Key
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Describing Chemical Reactions Study Guide Answer Key
 - User-Friendly Interface
4. Exploring eBook Recommendations from Describing Chemical Reactions Study Guide Answer Key
 - Personalized Recommendations
 - Describing Chemical Reactions Study Guide Answer Key User Reviews and Ratings
 - Describing Chemical Reactions Study Guide Answer Key and Bestseller Lists

5. Accessing Describing Chemical Reactions Study Guide Answer Key Free and Paid eBooks
 - Describing Chemical Reactions Study Guide Answer Key Public Domain eBooks
 - Describing Chemical Reactions Study Guide Answer Key eBook Subscription Services
 - Describing Chemical Reactions Study Guide Answer Key Budget-Friendly Options
6. Navigating Describing Chemical Reactions Study Guide Answer Key eBook Formats
 - ePub, PDF, MOBI, and More
 - Describing Chemical Reactions Study Guide Answer Key Compatibility with Devices
 - Describing Chemical Reactions Study Guide Answer Key Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Describing Chemical Reactions Study Guide Answer Key
 - Highlighting and Note-Taking Describing Chemical Reactions Study Guide Answer Key
 - Interactive Elements Describing Chemical Reactions Study Guide Answer Key
8. Staying Engaged with Describing Chemical Reactions Study Guide Answer Key
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Describing Chemical Reactions Study Guide Answer Key
9. Balancing eBooks and Physical Books Describing Chemical Reactions Study Guide Answer Key
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Describing Chemical Reactions Study Guide Answer Key
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Describing Chemical Reactions Study Guide Answer Key
 - Setting Reading Goals Describing Chemical Reactions Study Guide Answer Key
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Describing Chemical Reactions Study Guide Answer Key
 - Fact-Checking eBook Content of Describing Chemical Reactions Study Guide Answer Key
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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