#### **Balancing Act**

Name

Atoms are not <u>created</u> or <u>destroyed</u> during a chemical reaction.

Scientists know that there must be the <u>same</u> number of atoms on each <u>side</u> of the <u>reaction</u>. To balance the chemical equation, you must add <u>coefficients</u> in front of the chemical formulas in the equation. You cannot <u>add</u> or <u>remove</u> subscripts.

- 1) Determine number of atoms for each element
- 2) Pick an element that is not equal on both sides of the equation.
- 3) Add coefficients in front of the element and adjust your counts.
- Continue adding coefficients until you get the same number of elements on each side.

#### Example:

$$2$$
 Mg +  $O_2$   $\rightarrow$   $2$  MgO  
Mg =  $2$  Mg =  $2$   
O =  $2$  O =  $2$ 

#### Try these:

$$N_2$$
 + 3  $H_2$   $\rightarrow$  2  $NH_3$   
 $N = 2$   
 $H = 6$   $H = 6$ 

$$2 H_2O_2 \rightarrow 2 H_2O + O_2$$
 $H = 4 H = 4$ 
 $O = 4 O = 4$ 

# **Balancing Act Science Answer Key**

M Mark

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#### Balancing Act Science Answer Key Book Review: Unveiling the Magic of Language

In an electronic era where connections and knowledge reign supreme, the enchanting power of language has are more apparent than ever. Its capability to stir emotions, provoke thought, and instigate transformation is actually remarkable. This extraordinary book, aptly titled "**Balancing Act Science Answer Key**," written by a very acclaimed author, immerses readers in a captivating exploration of the significance of language and its profound affect our existence. Throughout this critique, we will delve in to the book is central themes, evaluate its unique writing style, and assess its overall influence on its readership.

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