Name: Hour: Date:

Chemistry: Balancing & Classifying Chemical Equations

<u>Directions</u>: First, balance each of the chemical equations below. Then, classify each reaction as **synthesis**, **decomposition**, **single-replacement**, or **double-replacement**. To earn full credit, write the words out when classifying.

Balance the equation...

...and classify it.

10. ___ Pb(NO₃)₂ + ___ H₂S
$$\rightarrow$$
 ___ PbS + ___ HNO₃

19. ___ Al + ___
$$H_2SO_4 \rightarrow$$
 ___ $Al_2(SO_4)_3 + ___ H_2$

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