

- _____ 1. A chemical reaction is a process in which
 a. products change into reactants c. substances change state
 b. the law of conservation of mass applies d. all of these
- _____ 2. During a chemical reaction,
 a. new elements are produced c. atoms are destroyed
 b. atoms are rearranged d. elements are destroyed
- _____ 3. An equation is balanced by
 a. changing subscripts c. erasing elements as necessary
 b. adding coefficients d. adding elements as necessary
- _____ 4. An atom's ability to undergo chemical reactions is determined by its
 a. protons b. innermost electrons c. neutrons d. outermost electrons
- _____ 5. What are the reactants in the following chemical equation:

$$\text{Zn} + \text{CuSO}_4 \longrightarrow \text{ZnSO}_4 + \text{Cu}$$
 a. zinc and copper c. zinc and copper (II) sulfate
 b. zinc sulfate and copper d. only zinc
- _____ 6. What are the products in the above equation?
 a. zinc and copper c. zinc and copper (II) sulfate
 b. zinc sulfate and copper d. only zinc

For questions 7-12, classify the reaction according to the type it is. Put that answer in the blank. Then add coefficients to balance the reaction when necessary.



Balancing Chemical Equations Multiple Choice Answer

Paul Foglino, Princeton Review



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