

# INTERMOLECULAR FORCES REVIEW


## INTERMOLECULAR

**H<sub>2</sub>O**



Bonding Electrons	Nonbonding Electrons	Total Electrons
2	2	4
VSEPR Pairs	VSEPR Angles	
Linear	180°	
Molecular Polarity	Intermolecular Forces	
<input checked="" type="checkbox"/> Polar	<input checked="" type="checkbox"/> Dipole	
<input type="checkbox"/> Nonpolar	<input type="checkbox"/> H-O	
	<input type="checkbox"/> H-H	

**CCl<sub>4</sub>**



Bonding Electrons	Nonbonding Electrons	Total Electrons
4	0	4
VSEPR Pairs	VSEPR Angles	
Tetrahedral	109.5°	
Molecular Polarity	Intermolecular Forces	
<input type="checkbox"/> Polar	<input checked="" type="checkbox"/> Dipole	
<input checked="" type="checkbox"/> Nonpolar	<input type="checkbox"/> H-O	
	<input type="checkbox"/> H-H	

## FORCES REVIEW

**CH<sub>4</sub>**

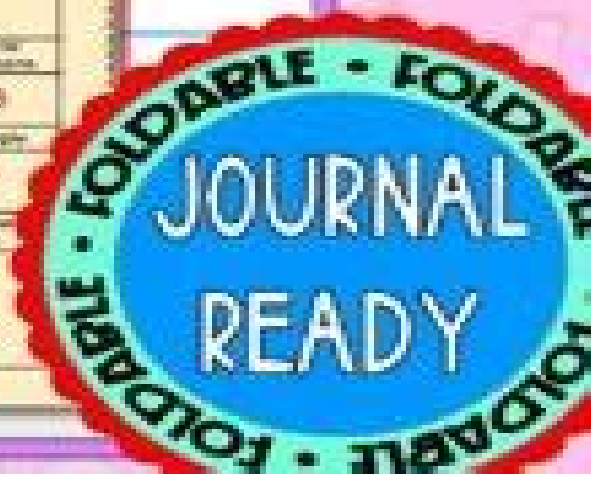


Bonding Electrons	Nonbonding Electrons	Total Electrons
2	0	2
VSEPR Pairs	VSEPR Angles	
Linear	180°	
Molecular Polarity	Intermolecular Forces	
<input type="checkbox"/> Polar	<input checked="" type="checkbox"/> Dipole	
<input checked="" type="checkbox"/> Nonpolar	<input type="checkbox"/> H-O	
	<input type="checkbox"/> H-H	

**H<sub>2</sub>O**



Bonding Electrons	Nonbonding Electrons	Total Electrons
2	0	2
VSEPR Pairs	VSEPR Angles	
Trigonal planar	120°	
Molecular Polarity	Intermolecular Forces	
<input checked="" type="checkbox"/> Polar	<input checked="" type="checkbox"/> Dipole	
<input type="checkbox"/> Nonpolar	<input type="checkbox"/> H-O	
	<input type="checkbox"/> H-H	



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**R. Thomas Myers**



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