

# 9 • Bonding & Molecular Structure

## LEWIS STRUCTURE PRACTICE - 1

$\text{Ca}^{2+}$ $[Ca]^{2+}$	$\text{NO}_3^-$ $\leftrightarrow \left[ \begin{array}{c} \text{O}^- - \text{N} = \text{O} \\   \\ \text{O}^- \end{array} \right]^-$	$\text{PCl}_5$ 
$\text{C}_2\text{H}_2$ $\text{H} : \text{C} : : : \text{C} : \text{H}$	$\text{S}^{2-}$ $[\text{:}\ddot{\text{S}}\text{:}]^{2-}$	$\text{NH}_3$ $\text{H} : \ddot{\text{N}} : \text{H}$ $ $ $\text{H}$
$\text{SO}_4^{2-}$ $\left[ \begin{array}{c} \text{O} \\ \text{:}\ddot{\text{O}}\text{:} \\ \text{:}\ddot{\text{S}}\text{:} \\ \text{:}\ddot{\text{O}}\text{:} \\ \text{O} \end{array} \right]^{2-}$	$\text{SO}_2$ $\text{O}^- - \text{S} = \text{O} \leftrightarrow \text{O} = \text{S} - \text{O}^-$	$\text{CaH}_2$ $\text{H} : \text{Ca} : \text{H}$



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## LEWIS STRUCTURE PRACTICE - 2

$\text{CH}_4$ $\text{H} - \text{C} - \text{H}$ $ $ $\text{H}$	$\text{HCN}$ $\text{H} : \text{C} : : : \text{N} :$	$\text{K}^+$ $[\text{K}]^+$
$\text{PO}_4^{3-}$ $\left[ \begin{array}{c} \text{O} \\ \text{:}\ddot{\text{O}}\text{:} \\ \text{:}\ddot{\text{P}}\text{:} \\ \text{:}\ddot{\text{O}}\text{:} \\ \text{O} \end{array} \right]^{3-}$	$\text{O}_3$ $\text{O}^- - \text{O} = \text{O} \leftrightarrow \text{O} = \text{O} - \text{O}^-$	$\text{SF}_6$ 
$\text{Br}^-$ $[\text{:}\ddot{\text{Br}}\text{:}]^-$	$\text{AlH}_3$ $\text{H} : \text{Al} : \text{H}$ $ $ $\text{H}$	$\text{NH}_4^+$ $\left[ \begin{array}{c} \text{H} \\ \text{H} : \text{N} : \text{H} \\   \\ \text{H} \end{array} \right]^+$

# Bonding And Molecular Structure Worksheet Answers

**Jicheng Xie**



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