

Q.1: When acid reacts with metal carbonate, products are

- A. salt
- B. water
- C. carbon dioxide
- D. all of above

Answer: Option D

Q.2: A substance that donates a pair of electrons to form coordinate covalent bond is called

- A. Lewis acid
- B. Lewis base
- C. Bronsted-Lowry acid
- D. Bronsted-Lowry base

Answer: Option B

Q.3: Process in which acids (H^+) and bases (OH^-) react to form salts and water is called

- A. neutralization
- B. hydrogenation
- C. halogenation
- D. sublimation

Answer: Option A

Q.4: Phenolphthalein in acidic solution is

- A. colorless
- B. pink colored
- C. yellow colored
- D. orange colored

Answer: Option A

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