## Question 4

Classify each of the following chemical reactions:

a) 
$$3 \operatorname{Sn}(s) + 2 \operatorname{P}(s) \rightarrow \operatorname{Sn}_3 \operatorname{P}_2(s)$$

combination

neutralization

c) 
$$Cu(s) + 2 AgC_2H_3O_2(aq) \rightarrow Cu(C_2H_3O_2)_2(aq) + 2 Ag(s)$$

[Select]

decomposition

double-replacement.

# **Classify Chemical Reactions Answer**

Nancy Wentworth, Kenneth J. Plummer, Richard H. Swan

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