Identifying Reaction Types & Balancing Equations

Step 1: Name the type of chemical reaction	Step 2: Balance the following chemical equations:
1. Ex: Synthesis	___ N_2 + ___ H_2 → ___ NH_3
2.	___ $KClO_3$ → ___ KCl + ___ O_2
3.	___ NaCl + ___ F_2 → ___ NaF + ___ Cl_2
4.	___ H_2 + ___ O_2 → ___ H_2O
5.	___ $AgNO_3$ + ___ $MgCl_2$ → ___ AgCl + ___ $Mg(NO_3)_2$
6.	___ $AlBr_3$ + ___ K_2SO_4 → ___ KBr + ___ $Al_2(SO_4)_3$
7.	___ CH_4 + ___ O_2 → ___ CO_2 + ___ H_2O
8.	___ C_3H_8 + ___ O_2 → ___ CO_2 + ___ H_2O
9.	___ $FeCl_3$ + ___ NaOH → ___ $Fe(OH)_3$ + ___ NaCl
10.	___ P + ___ O_2 → ___ P_2O_5
11.	___ Na + ___ H_2O → ___ NaOH + ___ H_2
12.	___ Ag_2O → ___ Ag + ___ O_2
13.	___ CO_2 + ___ H_2O → ___ $C_6H_{12}O_6$ + ___ O_2
14.	___ K + ___ $MgBr_2$ → ___ KBr + ___ Mg
15.	___ $HNO_3(aq)$ + ___ $Ba(OH)_2(aq)$ → ___ $Ba(NO_3)_2(aq)$ + ___ H_2O (l)
16.	___ $C_5H_{12}(g)$ + ___ $O_2(g)$ → ___ $CO_2(g)$ + ___ $H_2O(g)$
17.	___ Al(s) + ___ $Fe_2O_3(aq)$ → ___ $Al_2O_3(aq)$ + ___ Fe(s)
18.	___ Al(s) + ___ $O_2(g)$ → ___ Al_2O_3 (s)

Balancing Equations And Identifying Reaction Types Answer Key

Nivaldo J. Tro

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