Name: \_\_\_\_\_

Date: \_\_\_\_\_

## **Balancing Chemical Equations**

Balance the following chemical equations.

1. 
$$\underline{2}$$
 Fe +  $\underline{3}$  H<sub>2</sub>SO<sub>4</sub>  $\longrightarrow$   $\underline{1}$  Fe<sub>2</sub>(SO<sub>4</sub>)<sub>3</sub> +  $\underline{3}$  H<sub>2</sub>

2. 
$$\underline{1}$$
 CH<sub>4</sub> +  $\underline{2}$  O<sub>2</sub>  $\longrightarrow$   $\underline{1}$  CO<sub>2</sub> +  $\underline{2}$  H<sub>2</sub>O

3. 
$$1$$
 SiCl<sub>4</sub>( $\ell$ ) +  $2$  H<sub>2</sub>O( $\ell$ )  $\longrightarrow$   $1$  SiO<sub>2</sub>( $s$ ) +  $4$  HCl(aq)

4. 
$$\underline{2}$$
 AgI +  $\underline{1}$  Na<sub>2</sub>S  $\longrightarrow$   $\underline{1}$  Ag<sub>2</sub>S +  $\underline{2}$  NaI

5. 
$$\underline{4}$$
 NH<sub>3</sub> +  $\underline{5}$  O<sub>2</sub>  $\longrightarrow$   $\underline{4}$  NO +  $\underline{6}$  H<sub>2</sub>O

6. 1 FeO<sub>3</sub>(s) + 3 CO(g) 
$$\longrightarrow$$
 1 Fe( $\ell$ ) + 3 CO<sub>2</sub>(g)

7. 
$$\underline{1}$$
 SiO<sub>2</sub> +  $\underline{4}$  HF  $\longrightarrow$   $\underline{1}$  SiF<sub>4</sub> +  $\underline{2}$  H<sub>2</sub>O

9. 
$$4 (NH_4)_3PO_4 + 3 Pb(NO_3)_4 \longrightarrow 1 Pb_3(PO_4)_4 + 12 NH_4NO_3$$

# **Balancing Equations Practice Problems Work Answer Key**

John S. Phillips, Cheryl Wistrom

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# **Table of Contents Balancing Equations Practice Problems Work Answer Key**

- 1. Understanding the eBook Balancing Equations Practice Problems Work Answer Key
  - The Rise of Digital Reading Balancing Equations Practice Problems Work Answer Key
  - Advantages of eBooks Over Traditional Books
- 2. Identifying Balancing Equations Practice Problems Work Answer Key
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
- 3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Balancing Equations Practice Problems Work Answer Key
  - User-Friendly Interface
- 4. Exploring eBook Recommendations from Balancing Equations Practice Problems Work Answer Key
  - Personalized Recommendations
  - Balancing Equations Practice Problems Work Answer Key User Reviews and Ratings
  - Balancing Equations Practice Problems Work Answer Key and Bestseller Lists

- 5. Accessing Balancing Equations Practice Problems Work Answer Key Free and Paid eBooks
  - o Balancing Equations Practice Problems Work Answer Key Public Domain eBooks
  - Balancing Equations Practice Problems Work Answer Key eBook Subscription Services
  - Balancing Equations Practice Problems Work Answer Key Budget-Friendly Options
- 6. Navigating Balancing Equations Practice Problems Work Answer Key eBook Formats
  - o ePub, PDF, MOBI, and More
  - Balancing Equations Practice Problems Work Answer Key Compatibility with Devices
  - Balancing Equations Practice Problems Work Answer Key Enhanced eBook Features
- 7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Balancing Equations Practice Problems Work Answer Key
  - Highlighting and Note-Taking Balancing Equations Practice Problems Work Answer Key
  - Interactive Elements Balancing Equations Practice Problems Work Answer Key
- 8. Staying Engaged with Balancing Equations Practice Problems Work Answer Key
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Balancing Equations Practice Problems Work Answer Key
- 9. Balancing eBooks and Physical Books Balancing Equations Practice Problems Work Answer Key
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Balancing Equations Practice Problems Work Answer Key
- 10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
- 11. Cultivating a Reading Routine Balancing Equations Practice Problems Work Answer Key
  - Setting Reading Goals Balancing Equations Practice Problems Work Answer Key
  - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Balancing Equations Practice Problems Work Answer Key
  - Fact-Checking eBook Content of Balancing Equations Practice Problems Work Answer Key
  - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
- Exploring Educational eBooks
- 14. Embracing eBook Trends
  - Integration of Multimedia Elements
  - Interactive and Gamified eBooks

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