

Chapter 7 Study Guide

Foundations of Chemistry

Vocab to know

Matter	Compound	Physical property	Density	Chemical property
Atom	Mixture	Mass	Solubility	Chemical change
Substance	Homogeneous mixture	Heterogeneous mixture	Bond	Concentration
Element	Physical change	Dissolve		

Be able to answer the following questions:

- 1. What is a substance?**
- 2. How do atoms of different elements differ?**
- 3. What does a chemical formula show?**
- 4. How do substances combine in a mixture?**
- 5. How do mixtures differ from substances?**
- 6. How is a compound different from a solution?**
- 7. How can you classify matter?**
- 8. How do solids, liquids, and gases differ?**
- 9. What are five different physical properties of matter?**
- 10. What are two types of conductivity?**
- 11. How are physical properties used to separate mixtures?**
- 12. How can removing thermal energy affect the state of matter?**
- 13. What happens when something dissolves?**
- 14. What are some chemical properties of matter?**
- 15. What are the signs of a chemical change?**
- 16. Why are chemical equations useful?**
- 17. How do you know that a chemical equation is balanced?**
- 18. List 3 factors that affect the rate of a chemical reaction.**
- 19. How do the properties of the reactants in a chemical reaction compare to the properties of the products of the reaction?**

Century 2accounting Chapter 7 Study Guide Answers

David Floyd



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web 342 1 subject to subsection 2 a day reporting order may in addition include requirements for securing the electronic

monitoring of the whereabouts of an offender subject to the order during the period when the order is in force against the offender 2 a court must not make a day reporting order which includes the requirements