# BALANCING EQUATIONS: FORMULAS GIVEN

# Practice Sheet #1

# Balance the following equations:

- Al + N<sub>1</sub> → AlN
- 2. Fe + Q  $\rightarrow$  Fe<sub>2</sub>O<sub>4</sub>
- 3.  $CaCO_2 \rightarrow CaO + CQ$
- 4. NHNO<sub>1</sub> → NO + HO
- KI + Cl<sub>2</sub> → KCl + I
- Pb(NO<sub>1</sub>)<sub>2</sub>+ HCl → PbCl<sub>2</sub> + HNO<sub>1</sub>
- 7.  $BaO_2 \rightarrow BaO + Q$
- 8. Al +  $H_sSO_s \rightarrow Al(SO_s)_1 + H_s$
- CH<sub>4</sub> + Cl<sub>2</sub> → CHCl<sub>3</sub> + HCl
- MgCl<sub>2</sub> + NaOH → Mg(OH) + NaCl
- 11.  $AgNO_1 + CuCl_2 \rightarrow AgCl + Cu(NO_1)_2$
- 12.  $ZnS + Q \rightarrow ZnO + SQ$
- 13. Na + HO → H + NaOH
- BaCl<sub>2</sub> + (NH)<sub>2</sub>CO<sub>3</sub> → BaCO<sub>4</sub> + NH<sub>6</sub>CI
- 15.  $C_nH_n + O_2 \rightarrow CO_2 + H_0O$
- 16. Na + HO → NoH + H
- 17. Fe + FeCl → FeCl
- 18.  $Ba(OH)_1 + AlCl_2 \rightarrow Al(OH) + BaCl_2$
- H<sub>2</sub>C<sub>2</sub>O<sub>4</sub> + KOH → KC<sub>2</sub>O<sub>4</sub> + H<sub>2</sub>O
- 20.  $C_2H_2Cl_4 + C_4(OH) \rightarrow C_2HCl_2 + C_4Cl_2 + H_2O$
- 21.  $(NH_1)_1Cr_2O_7 \rightarrow N_1 + Cr_2O_3 + H_2O_4$

# **Balancing Equations Formulas Given Answers**

Wenbin Ji

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mind they have anticipated where students might stumble and have paced and organized this text to help them through Their goal is to make the material interesting and relevant so students understand the basic chemical principles related to their career The authors emphasize problem solving and provide a range of practice exercises As in previous editions the text first presents the basic concepts of general chemistry and then moves into organic and biochemistry In this edition the first two sections have been revised primarily to improve explanations and include new pedagogical features The biochemistry portion has been thoroughly updated to include coverage of many recent developments and emerging technologies in the field

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