

CHAPTER 7: GASES, LIQUIDS, AND SOLIDS

CHE 120

CHAPTER 07 GASES, LIQUIDS and SOLIDS Practice Questions

- Which of the following statements is not a part of kinetic molecular theory?
 - Matter is composed of particles that are in constant motion.
 - Particle velocity increases as the temperature increases.
 - Particles in a system cannot transfer energy to each other.
 - Particle interactions involve electrostatic attractions and repulsions.
- In which of the following groupings are the three terms closely related?
 - kinetic energy, energy of motion, cohesive forces
 - potential energy, energy of attraction, disruptive forces
 - kinetic energy, electrostatic interactions, disruptive forces
 - potential energy, electrostatic interactions, cohesive forces
- In the liquid state, disruptive forces are
 - roughly of the same magnitude as cohesive forces
 - very weak compared to cohesive forces
 - dominant over cohesive forces
 - unimportant
- The phrases "particles close together and held in fixed positions" and "completely fills the container" apply, respectively, to
 - liquids and solids
 - solids and gases
 - gases and liquids
 - liquids and gases
- How many times larger in size is the atmosphere pressure unit (atm) than the mm Hg pressure unit?
 - 100
 - 325
 - 760
 - 1000
- Boyle's law involves which of the following?
 - a direct proportion
 - a constant volume
 - a constant pressure
 - a constant temperature
- Indicate what the missing words are in the following statement of Charles's law: At constant pressure, the volume of a gas sample is _____ proportional to its _____ temperature.
 - directly; Celsius
 - directly; Kelvin
 - inversely; Celsius
 - inversely; Kelvin

Chapter Solids Liquids And Gases Test

Roman Wölfel



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