

Section 10.1 Measuring Matter

Objectives

- **Explain** how a mole is used to indirectly count the number of particles of matter.
- **Relate** the mole to a common everyday counting unit.
- **Convert** between moles and number of representative particles.

MAIN Idea

Chemists use the mole to count atoms, molecules, ions, and formula units.

Review Vocabulary

molecule: two or more atoms that covalently bond together to form a unit

New Vocabulary

mole

Avogadro's number



Chapter Study Guide The Mole Section 10measuring Matter

Peter McPherson



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Practical Volumetric Analysis Peter McPherson, 2014-09-17 Written by someone who has experienced both teaching and working as a research chemist this textbook will provide the theoretical chemistry associated with volumetric analysis supported by a selection of practicals for undergraduate students taking modules in introductory and analytical chemistry as well as for non specialists teaching chemistry Chemistry, Student Study Guide James E. Brady, Fred Senese, 2008-01-28

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