

Chapter Test B

Chapter: Chemical Formulas and Chemical Compounds

PART I In the space provided, write the letter of the term or phrase that best completes each statement or best answers each question.

- _____ 1. Changing a subscript in a correctly written chemical formula will
- change the electron configuration of that element.
 - change the charges on the other ions in the compound.
 - change the formula so that it no longer represents the same compound.
 - have no effect on the formula.
- _____ 2. Using the Stock system of nomenclature, $\text{Cr}_2(\text{SO}_4)_3$ is named
- chromium(II) sulfate.
 - chromic sulfate.
 - dichromium trisulfate.
 - chromium(III) sulfate.
- _____ 3. In a polyatomic ion, the algebraic sum of the oxidation numbers of all atoms is equal to
- 0.
 - 10.
 - the number of atoms in the ion.
 - the charge on the ion.
- _____ 4. The first part of the name of a binary ionic compound is the
- cation.
 - polyatomic ion.
 - oxyanion.
 - anion.
- _____ 5. The empirical formula may not represent the actual composition of a(n)
- ionic compound.
 - crystal.
 - salt.
 - molecular compound.
- _____ 6. The formula for phosphoric acid is
- HPO_4 .
 - $\text{H}(\text{PO}_4)_2$.
 - H_2PO_3 .
 - H_3PO_4 .

Chapter Test B Chemical Formulas And Compounds

Wenbin Ji

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Chapter Test B Chemical Formulas And Compounds Book Review: Unveiling the Magic of Language

In an electronic era where connections and knowledge reign supreme, the enchanting power of language has are more apparent than ever. Its ability to stir emotions, provoke thought, and instigate transformation is really remarkable. This extraordinary book, aptly titled "**Chapter Test B Chemical Formulas And Compounds**," compiled by a very acclaimed author, immerses readers in a captivating exploration of the significance of language and its profound affect our existence. Throughout this critique, we shall delve in to the book is central themes, evaluate its unique writing style, and assess its overall influence on its readership.

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