

# Dry Lab 3 Report Sheet

## Atomic and Molecular Structure

Date \_\_\_\_\_ Lab Sec. \_\_\_\_\_ Name \_\_\_\_\_ Desk No. \_\_\_\_\_

### A. The Mercury Spectrum

Instructor's approval of the wavelength grid of the spectra on the color plate (back cover): Hydrogen

### B. The Spectra of Elements

1. Spectrum number \_\_\_\_\_ is the emission line spectrum for hydrogen on the color plate.

What are the wavelengths and colors of the emission lines of the visible spectrum of hydrogen?

Red: 656.2 nm

Blue: 486.1 nm

Violet: 434.0

Indigo: 410.0

### 2. Identification of Spectra

a. Spectrum number \_\_\_\_\_

Intense lines in the spectrum: \_\_\_\_\_ nm; \_\_\_\_\_ nm; \_\_\_\_\_ nm; \_\_\_\_\_ nm; \_\_\_\_\_ nm; \_\_\_\_\_ nm; \_\_\_\_\_ nm

Element producing the spectrum \_\_\_\_\_

b. Spectrum number \_\_\_\_\_

Intense lines in the spectrum: \_\_\_\_\_ nm; \_\_\_\_\_ nm; \_\_\_\_\_ nm; \_\_\_\_\_ nm; \_\_\_\_\_ nm; \_\_\_\_\_ nm; \_\_\_\_\_ nm

Element producing the spectrum \_\_\_\_\_

# Dry Lab 3 Atomic Molecular Structure Answers

**David Baud**



### Dry Lab 3 Atomic Molecular Structure Answers:

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