

Chapter 1-Chemical Reactions and Equations

Question 2: $\text{Fe}_2\text{O}_3 + 2\text{Al} \rightarrow \text{Al}_2\text{O}_3 + 2\text{Fe}$ The above reaction is an example of a

- (a) combination reaction
- (b) double displacement reaction
- (c) decomposition reaction
- (d) displacement reaction

Answer:

(d) Displacement reaction.

Question 3: What happens when dilute hydrochloric acid is added to iron filings ? Tick the correct answer :

- (a) Hydrogen gas and iron chloride are produced.
- (b) Chlorine gas and iron hydroxide are produced.
- (c) No reaction takes place.
- (d) Iron salt and water are produced.

Answer:

(a) Hydrogen gas and iron chloride are produced.

Question 4 :What is a balanced chemical equation ? Why should chemical equations be balanced ?

Answer:A balanced chemical equation has an equal number of atoms of different elements in the reactants and products.

The chemical equations should be balanced to satisfy the law of conservation of mass.

Ch 10 1 Reactions And Equations Answer Key

Léo Pomar



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