Name:		Date: Per	iodi
	Percentage (	Composition Worksheet	
Give the % compositio	n of all elements in th	ese compounds. Show all work!	
Remember: % co	mpositoin of e	$lement = \frac{mass\ of\ element}{mass\ of\ compund} \times 100$	
When calculating mass	of element don't for	get to account for the subscripts.	
<ol> <li>ammonium sulfite</li> </ol>	$(NH_4)_2SO_3$	molar mass of compound:	
% N			
% H			
% S			
% O			
2. aluminum acetate % Al	$Al(C_2H_3O_2)_3$	molar mass of compound:	
% C			
% H			
% O			
3. sodium bromide % Na	NaBr	molar mass of compound:	
% Br			
4. copper (II) hydrox % Cu	ide Cu(OH) <sub>2</sub>	molar mass of compound:	
% O			
% H			

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Robert C. Smoot, Smoot, Richard G. Smith, Jack Price

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