# 9

# **CHEMICAL NAMES AND FORMULAS**

# Chapter Test A

### A. Matching

Match each description in Column B to the correct term in Column A. Write the letter of the correct description in the blank.

	Column A		Column B
	. cation	а.	any atom or group of atoms with a negative charge
2	. anion	b.	When two elements combine to form more than one compound, the different masses of one element that combine with the same mass of the other element are in the ratio of small whole numbers.
3	Law of definite proportions	e.	tightly bound group of atoms that behaves as a unit and carries a charge
	. acid	d.	a compound that produces hydroxide ions when dissolved in water
5	. base	e.	a compound composed of two different elements
6	law of multiple proportions	r.	any atom or group of atoms with a positive charge
7	. polyatomic ion	g.	In all samples of the same chemical compound, the elements are always combined in the same proportion by mass.
8	L monatomic ion	h.	a compound that produces hydrogen ions when dissolved in water
9	. binary compound	36.	consists of a single atom with a positive or negative charge

#### **B. Multiple Choice**

Choose the best answer and write its letter on the line.

- \_\_\_\_\_\_10. The correct name for the N<sup>3-</sup> ion is the: a. nitrate ion.
  - b. nitric ion.

- c. nitride ion.
- d. nitrite ion.

# **Chapter 9 Chemical Names And Formulas Packet Answers**

Darrell D. Ebbing, Susan M. Schelble

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#### **Table of Contents Chapter 9 Chemical Names And Formulas Packet Answers**

- 1. Understanding the eBook Chapter 9 Chemical Names And Formulas Packet Answers
  - The Rise of Digital Reading Chapter 9 Chemical Names And Formulas Packet Answers
  - Advantages of eBooks Over Traditional Books
- 2. Identifying Chapter 9 Chemical Names And Formulas Packet Answers
  - Exploring Different Genres
  - o Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
- 3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Chapter 9 Chemical Names And Formulas Packet Answers
  - User-Friendly Interface
- 4. Exploring eBook Recommendations from Chapter 9 Chemical Names And Formulas Packet Answers
  - Personalized Recommendations
  - Chapter 9 Chemical Names And Formulas Packet Answers User Reviews and Ratings

- Chapter 9 Chemical Names And Formulas Packet Answers and Bestseller Lists
- 5. Accessing Chapter 9 Chemical Names And Formulas Packet Answers Free and Paid eBooks
  - Chapter 9 Chemical Names And Formulas Packet Answers Public Domain eBooks
  - o Chapter 9 Chemical Names And Formulas Packet Answers eBook Subscription Services
  - Chapter 9 Chemical Names And Formulas Packet Answers Budget-Friendly Options
- 6. Navigating Chapter 9 Chemical Names And Formulas Packet Answers eBook Formats
  - o ePub, PDF, MOBI, and More
  - Chapter 9 Chemical Names And Formulas Packet Answers Compatibility with Devices
  - Chapter 9 Chemical Names And Formulas Packet Answers Enhanced eBook Features
- 7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Chapter 9 Chemical Names And Formulas Packet Answers
  - Highlighting and Note-Taking Chapter 9 Chemical Names And Formulas Packet Answers
  - Interactive Elements Chapter 9 Chemical Names And Formulas Packet Answers
- 8. Staying Engaged with Chapter 9 Chemical Names And Formulas Packet Answers
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Chapter 9 Chemical Names And Formulas Packet Answers
- 9. Balancing eBooks and Physical Books Chapter 9 Chemical Names And Formulas Packet Answers
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Chapter 9 Chemical Names And Formulas Packet Answers
- 10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
- 11. Cultivating a Reading Routine Chapter 9 Chemical Names And Formulas Packet Answers
  - Setting Reading Goals Chapter 9 Chemical Names And Formulas Packet Answers
  - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Chapter 9 Chemical Names And Formulas Packet Answers
  - Fact-Checking eBook Content of Chapter 9 Chemical Names And Formulas Packet Answers
  - Distinguishing Credible Sources

- 13. Promoting Lifelong Learning
  - Utilizing eBooks for Skill Development
  - Exploring Educational eBooks
- 14. Embracing eBook Trends
  - Integration of Multimedia Elements
  - Interactive and Gamified eBooks

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